

# Experimental techniques in nuclear and particle physics

## (part 1)

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# Documentation and further reading

- Experimental techniques in Nuclear and particle physics, S. Tavernier, Springer, 2010
- Radiation detection and measurement, by Glen F. Knoll, 4rd edition, John Wiley & Sons, 2010
- Review of particle physics', Physics Letters (2010), This document is downloadable from the WEB at: <http://pdg.lbl.gov/pdg.html>.
- Nuclear Data Center tables, Brookhaven national laboratory, available from the WEB at <http://www.nndc.bnl.gov/>.
- Photomultiplier tubes, Hamamatsu, [http://sales.hamamatsu.com/assets/pdf/catsandguides/PMT\\_handbook\\_v3aE.pdf](http://sales.hamamatsu.com/assets/pdf/catsandguides/PMT_handbook_v3aE.pdf)

## A few words about symbols and units

energy : eV, keV, MeV, GeV, TeV, PeV;  $1\text{eV} = 1.602 \cdot 10^{-19} \text{ J}$

particle mass m:  $mc^2$

for photon  $E=Pe= \hbar\omega$

$\beta, \gamma$  : relativistic parameters Lorentz transformation

$\alpha$ : fine structure constant  $\approx 1/137$

$\hbar, c$  : not always written in equations

$r_0$  : classical electron radius  $\approx 2.8 \cdot 10^{-15} \text{ m}$

charge Z: as number of protons charges

radioactivity: bequerel Bq; 1 decay/s

absorbed dose: gray Gy; 1 J/kg

equivalent dose: sievert Sv

# Interactions of energetic subatomic particles in matter

# Subatomic particles?

The fundamental constituents of matter are quarks, leptons, and the quanta of force (gamma, W, Z, gluons). However, most of these you will never observe directly.

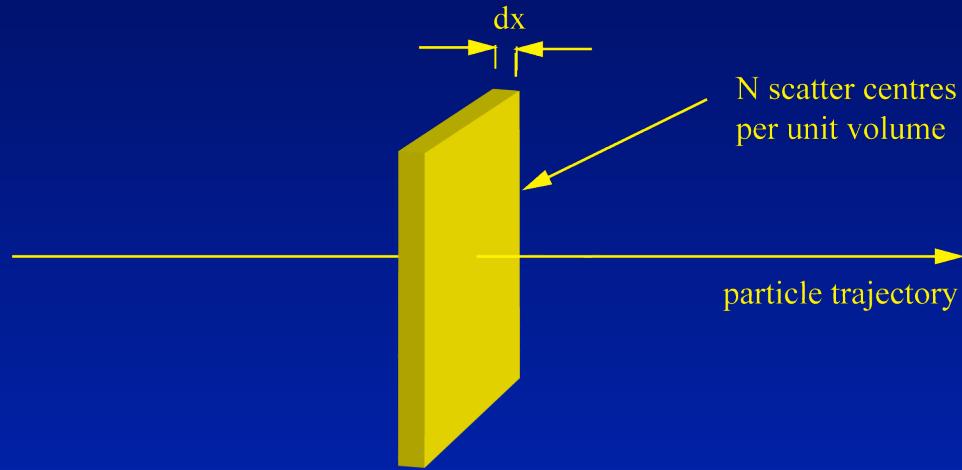
**Table 1.4: List of the most common directly observable particles.**

Particle	Mass	Lifetime	Charge	Main interactions
Electron	0.511 MeV	Stable	-1	Electromagnetic
Muon	105.7 MeV	$2.2 \cdot 10^{-6}$ s	-1	Electromagnetic
Hadrons	see table 1.1	see table 1.1	0, +1, -1	Strong
Photon	0	Stable	0	Electromagnetic
Nuclei	1–240, times the proton mass	Many stable	1–92	Strong
Neutrinos	$< \approx 2\text{eV}$	Stable	0	Weak

**Table 1.1: Some hadrons and their main properties. In this table the anti-particle of particle x is written as  $\bar{x}$ .**

Name	quark content	lifetime [s]	mass $\times c^2$ [MeV]	main decay modes
$\pi^+$	{u, $\bar{d}$ }	$2.6 \cdot 10^{-8}$	139.6	$\mu^+ \nu_\mu$
$\pi^-$	{d, $\bar{u}$ }	$2.6 \cdot 10^{-8}$	139.6	$\mu^- \bar{\nu}_\mu$
$\pi^0$	{u, $\bar{u}$ }, {d, $\bar{d}$ }	$1.2 \cdot 10^{-17}$	135.0	$2 \gamma$
$K^+$	{u, $\bar{s}$ }	$1.2 \cdot 10^{-8}$	473.7	$(\mu^+ \nu_\mu), (2\pi), (3\pi)$
$K^-$	{s, $\bar{u}$ }	$1.2 \cdot 10^{-8}$	473.7	$(\mu^-, \bar{\nu}_\mu), (2\pi), (3\pi)$
$K_l^o$	{s $\bar{d}$ }, {d $\bar{s}$ }	$5.2 \cdot 10^{-8}$	497.7	$3 \pi$
$K_s^0$	{s, $\bar{d}$ }, {d, $\bar{s}$ }	$8.9 \cdot 10^{-11}$	497.7	$2 \pi$
proton	{u, u, d}	$> 10^{32}$ year	938.3	
neutron	{d, d, u}	898	939.6	$p e^- \bar{\nu}_e$
$\Lambda^0$	{d, s, u}	$2.63 \cdot 10^{-10}$	1115.7	$(n \pi^0), (p \pi^-)$
$\bar{p}$	{ $\bar{u}$ , $\bar{u}$ , $\bar{d}$ }	$> 10^{32}$ year	938.3	
$\bar{n}$	{ $\bar{d}$ , $\bar{d}$ , $\bar{u}$ }	898	939.6	$\bar{p} e^+ \nu_e$
$\bar{\Lambda}^0$	{ $\bar{d}$ , $\bar{s}$ , $\bar{u}$ }	$2.63 \cdot 10^{-10}$	1115.7	$(\bar{n} \pi^0), (\bar{p} \pi^+)$

# Cross section $\sigma$ and mean free path $\lambda$



$$dW = dx N \sigma$$

$$1 \text{ barn} = 10^{-24} \text{ cm}^2$$

$$\text{Nucleus } N \rightarrow N = \frac{\rho N_{\text{Avogadro}}}{A_r}$$

$$\text{Mean free path} = \lambda = \frac{1}{N \sigma}$$

# Electromagnetic interactions of charged particles

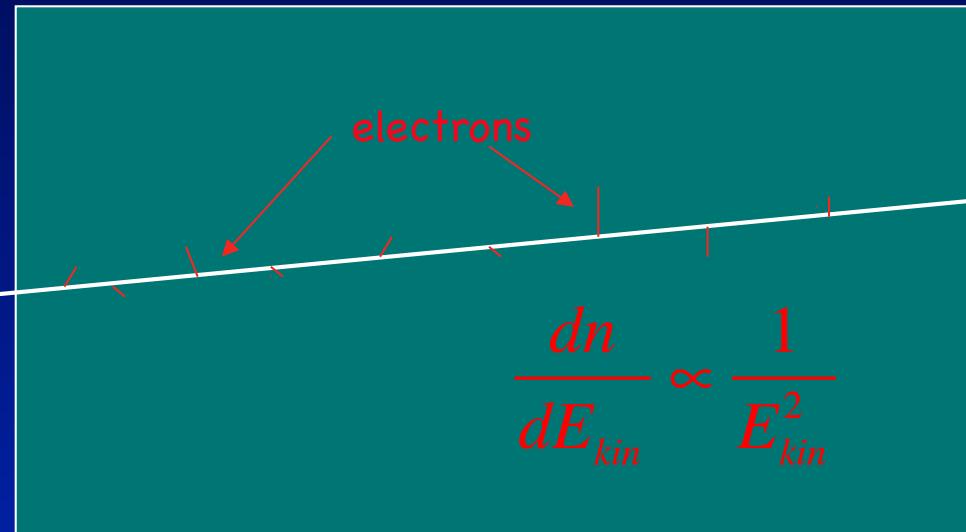
For a high energy particle, and to first approximation, matter can be seen as a collection of loosely bound electrons and nuclei

- collisions with nuclei
- collisions with electrons

# Collisions with electrons:

little change in direction, loss of energy

Charged particle



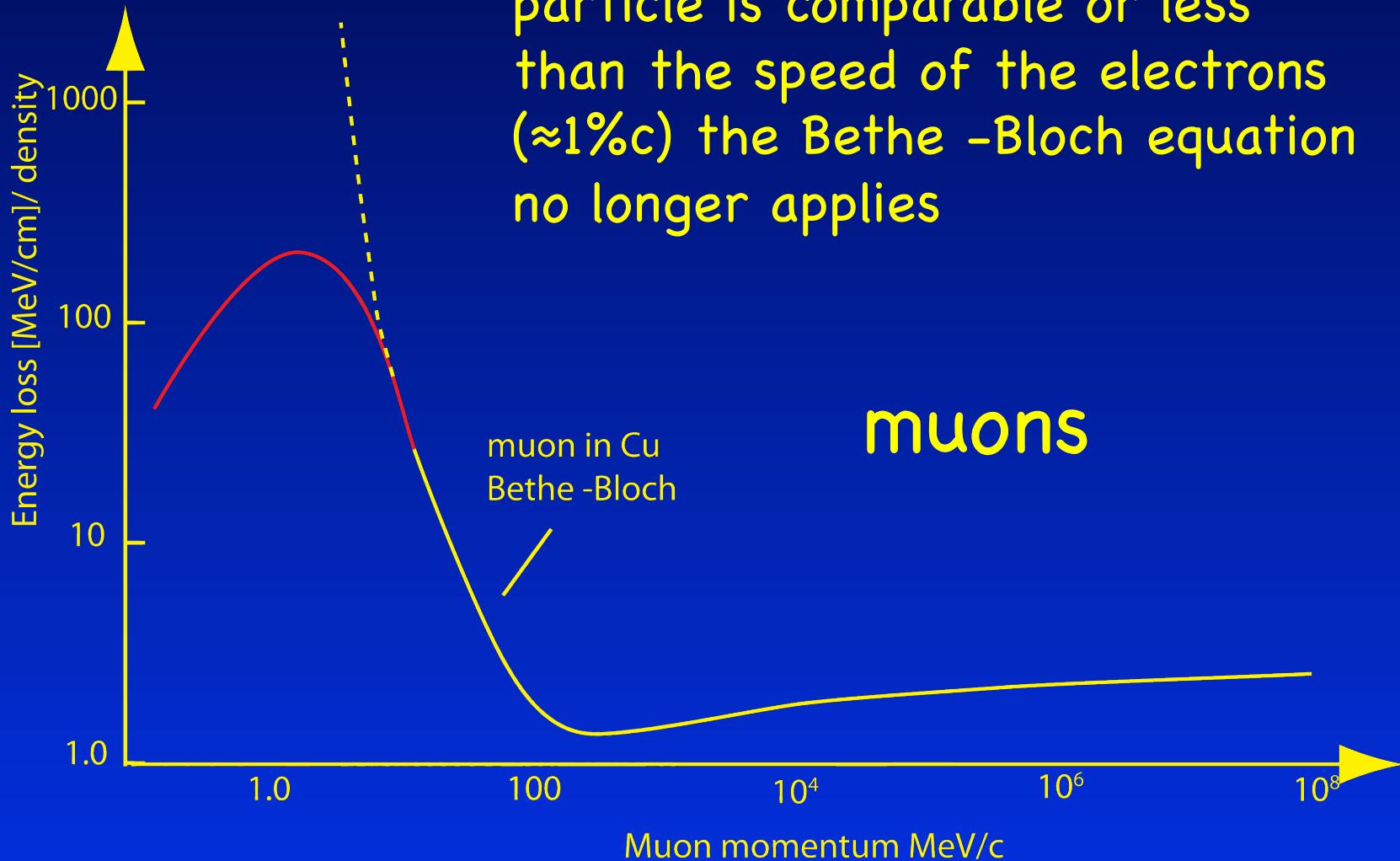
Bethe-Bloch equation for energy loss due to collisions with electrons

$$-\frac{dE}{dx} = \rho \frac{Z_{nucl}}{A_r} (4\pi N_A r_0^2 m_e c^2) \frac{Z^2}{\beta^2} \left[ \frac{1}{2} \ln \left( \frac{2m_e c^2 \beta^2 \gamma^2 T_{max}}{I^2} \right) - \beta^2 - \frac{\delta(\beta)}{2} \right]$$

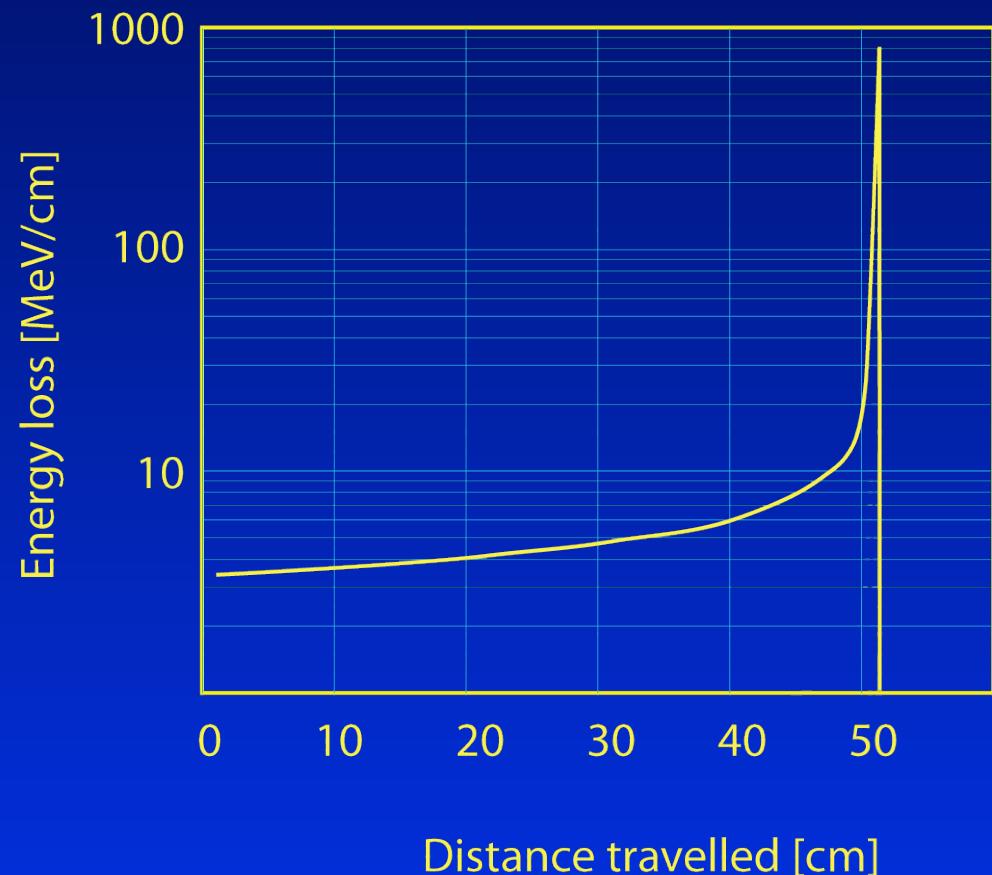
$$-\frac{dE}{dx} \approx \rho (2 \text{ MeV cm}^2) \frac{Z^2}{\beta^2}$$

for electrons similar

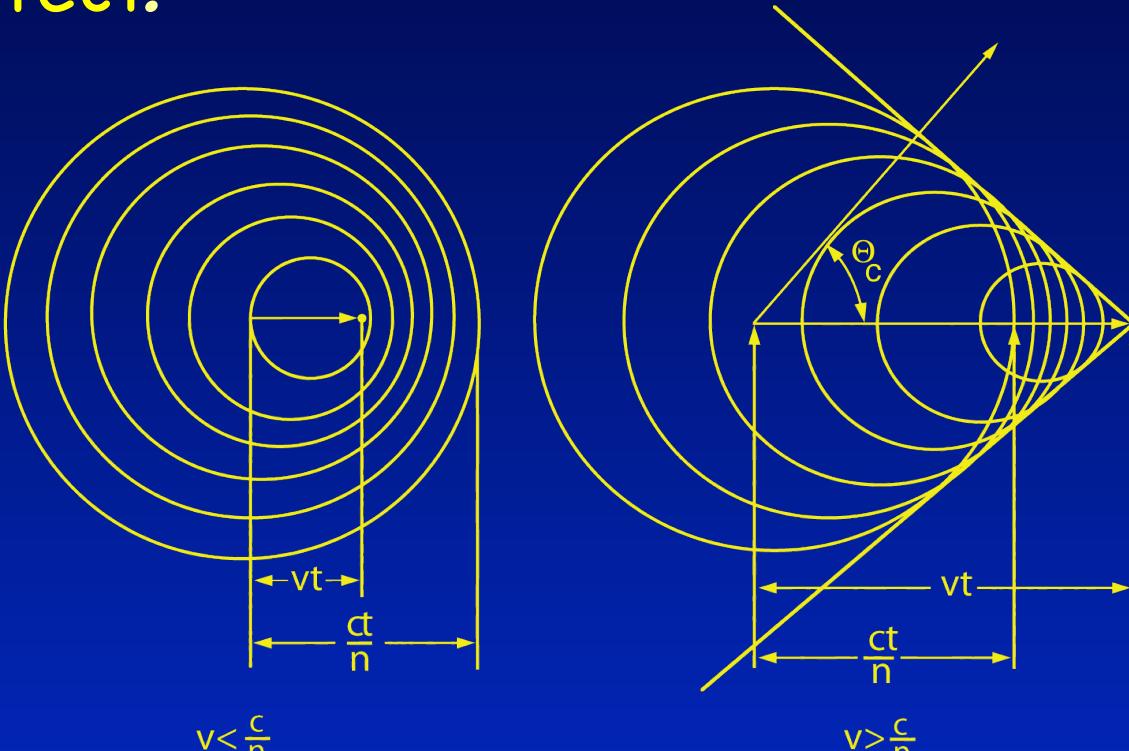
If the velocity of the charged particle is comparable or less than the speed of the electrons ( $\approx 1\%c$ ) the Bethe - Bloch equation no longer applies



# Range of a 300 MeV proton in water



# Cherenkov effect.



$$\frac{d^2E}{d\hbar\omega dx} = \hbar\omega \frac{Z^2\alpha}{\hbar c} \left[ 1 - \frac{c^2}{n^2v^2} \right]$$

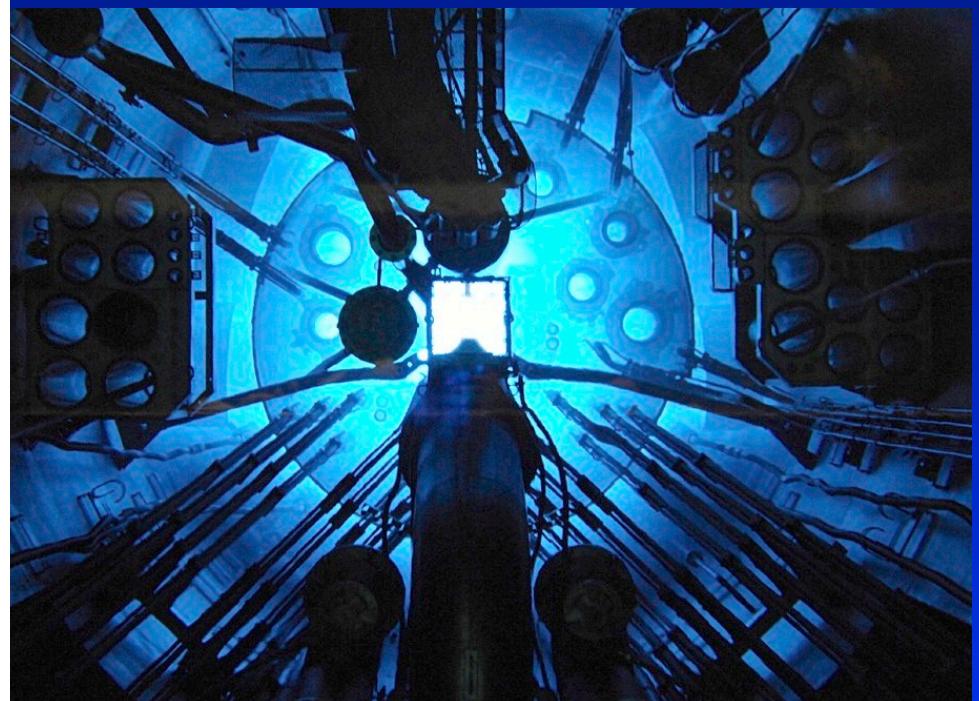
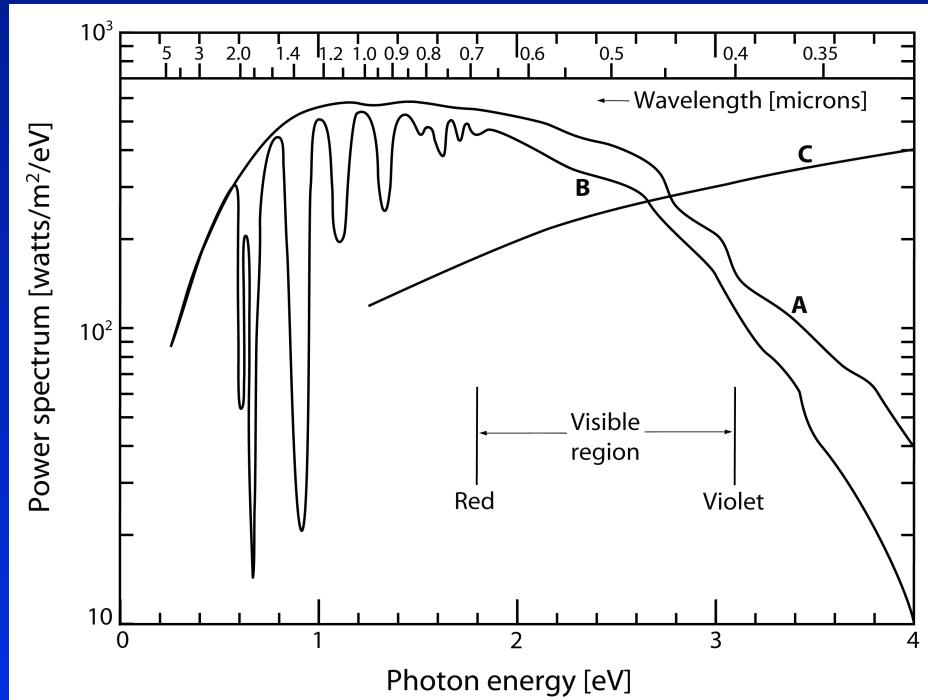
$$\cos(\theta_c) = \frac{c}{nv}$$

$$v > \frac{c}{n}$$

$$E_{threshold} = mc^2 \left( \sqrt{\frac{n^2}{n^2 - 1}} - 1 \right)$$

In water  $\approx 220$  photons/cm in visible (400-700 nm)  
about 100-1000 times less than a good scintillator!

Interest stems from 'v' dependence intensity and angle

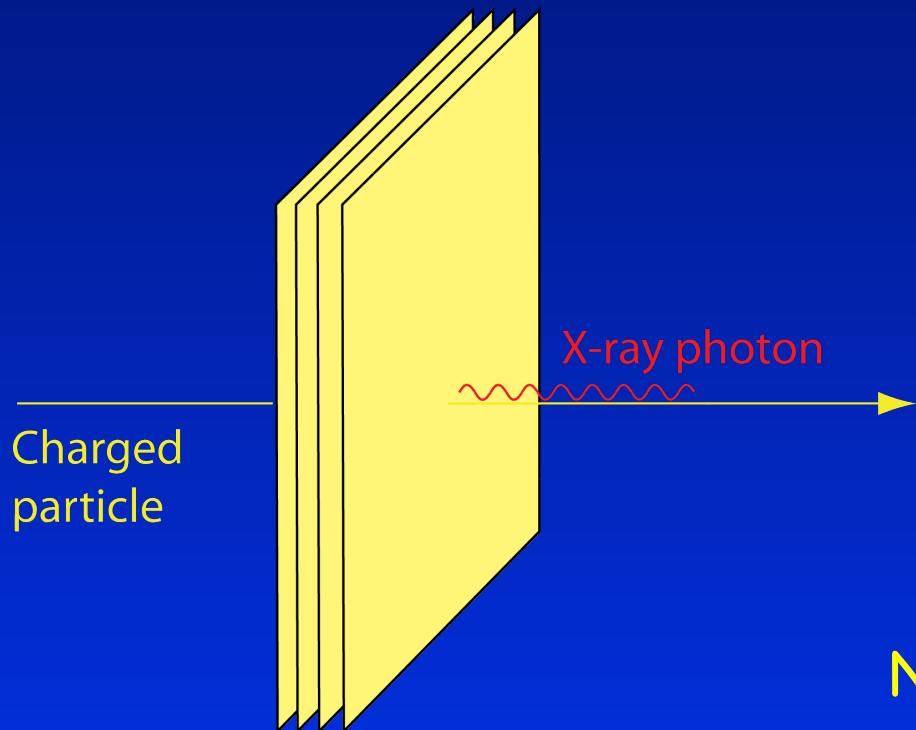


# Transition radiation

Plasma frequency  $\hbar\omega_p = \sqrt{4\pi N_e r_0^3} m_e c^2 / \alpha \approx \rho \times 20 \text{ eV}$

thin foils

$$\langle E \rangle = \alpha Z^2 \gamma \hbar \omega_p / 3$$



Number of photons  
with  $E > \gamma \hbar \omega_p / 10$

$$N_\gamma \simeq 0.8 \alpha Z^2 \simeq 0.59\% Z^2$$

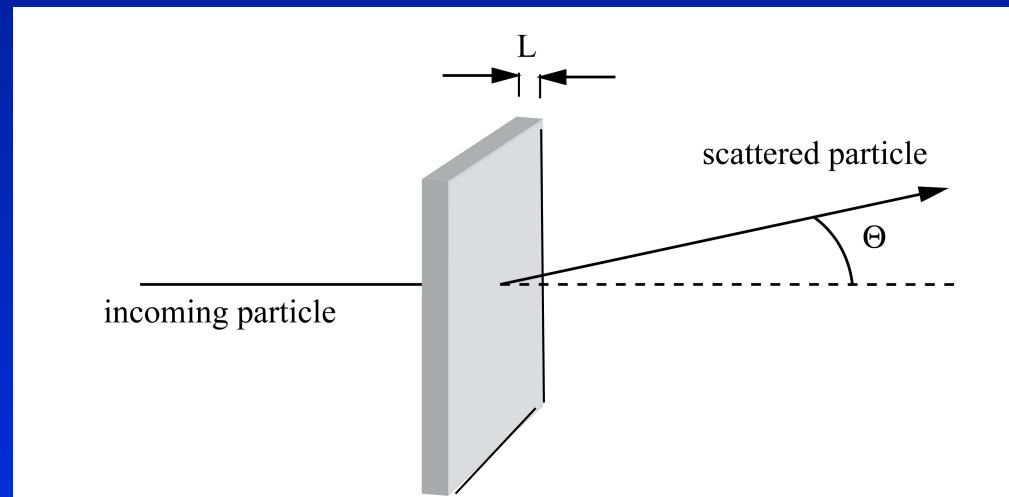
foils need thickness  $> \frac{\gamma c}{\omega_p}$

Notice the effects is  $\propto \gamma$   
and is significant only at large  $\gamma$

# Collisions with nuclei: little energy loss, change of direction

$$\sqrt{\langle \Theta^2 \rangle} = \frac{Z}{Pc\beta} (20 \text{ MeV}) \sqrt{\frac{L}{X_0}}$$

radiation length=  $X_0 = \frac{A_r \times 716.4 \text{ [cm}^{-2}\text{]}}{\rho Z_{nucl}(Z_{nucl} + 1) \ln\left(\frac{287}{\sqrt{Z_{nucl}}}\right)}$



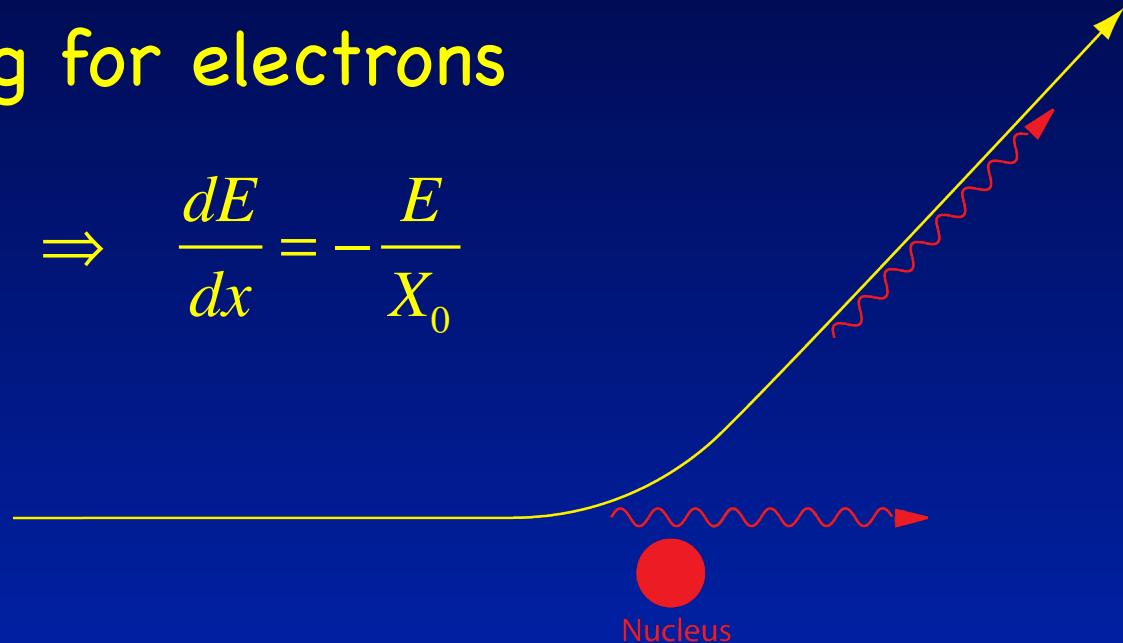
$$\sqrt{\langle \Theta_p \rangle^2} = \frac{1}{\sqrt{2}} \sqrt{\langle \Theta \rangle^2}$$

**Table 2.1: Radiation length  $X_0$  for some common materials**

Material	Radiation length $X_0$
Air	304 m
Water	36 cm
Shielding concrete	10.7 cm
Nylon	36.7 cm
Aluminium (Al)	8.9 cm
Silicon (Si)	9.36 cm
Iron (Fe)	1.76 cm
Lead (Pb)	0.56 cm
Uranium (U)	0.32 cm

## Bremsstrahlung for electrons

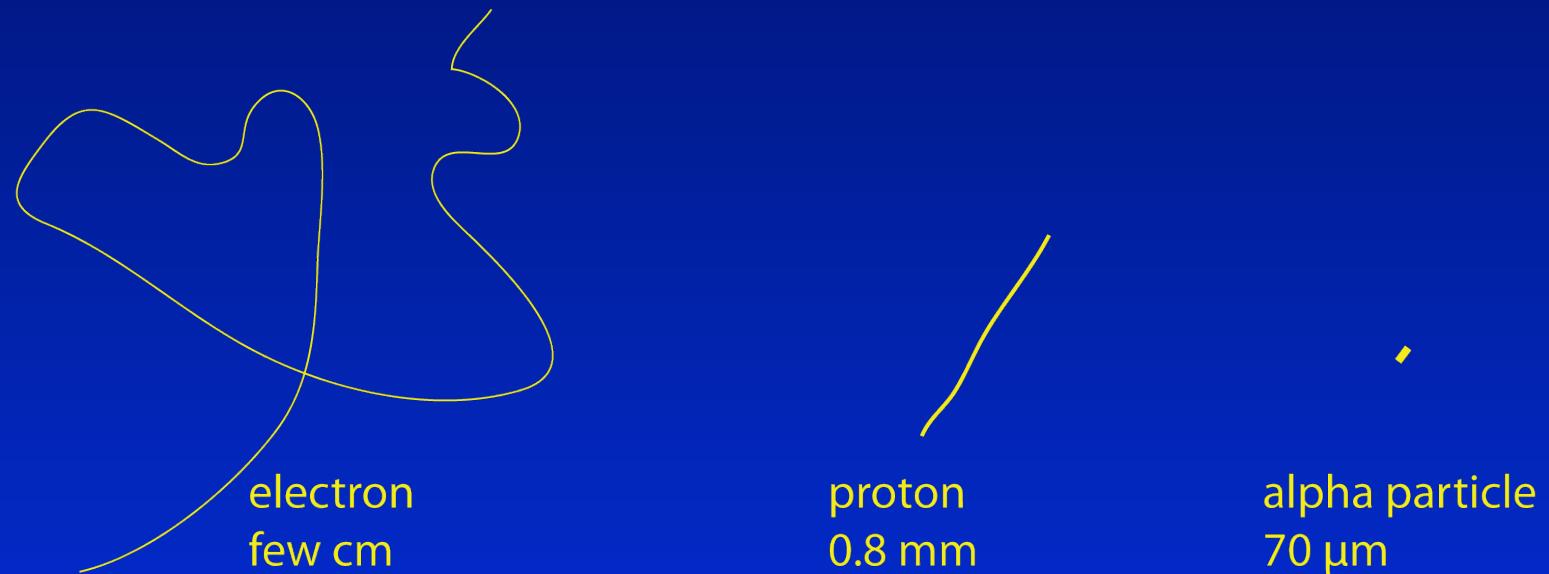
$$\frac{dN}{dx d\hbar\omega} = \frac{1}{\hbar\omega} \frac{1}{X_0} \Rightarrow \frac{dE}{dx} = -\frac{E}{X_0}$$

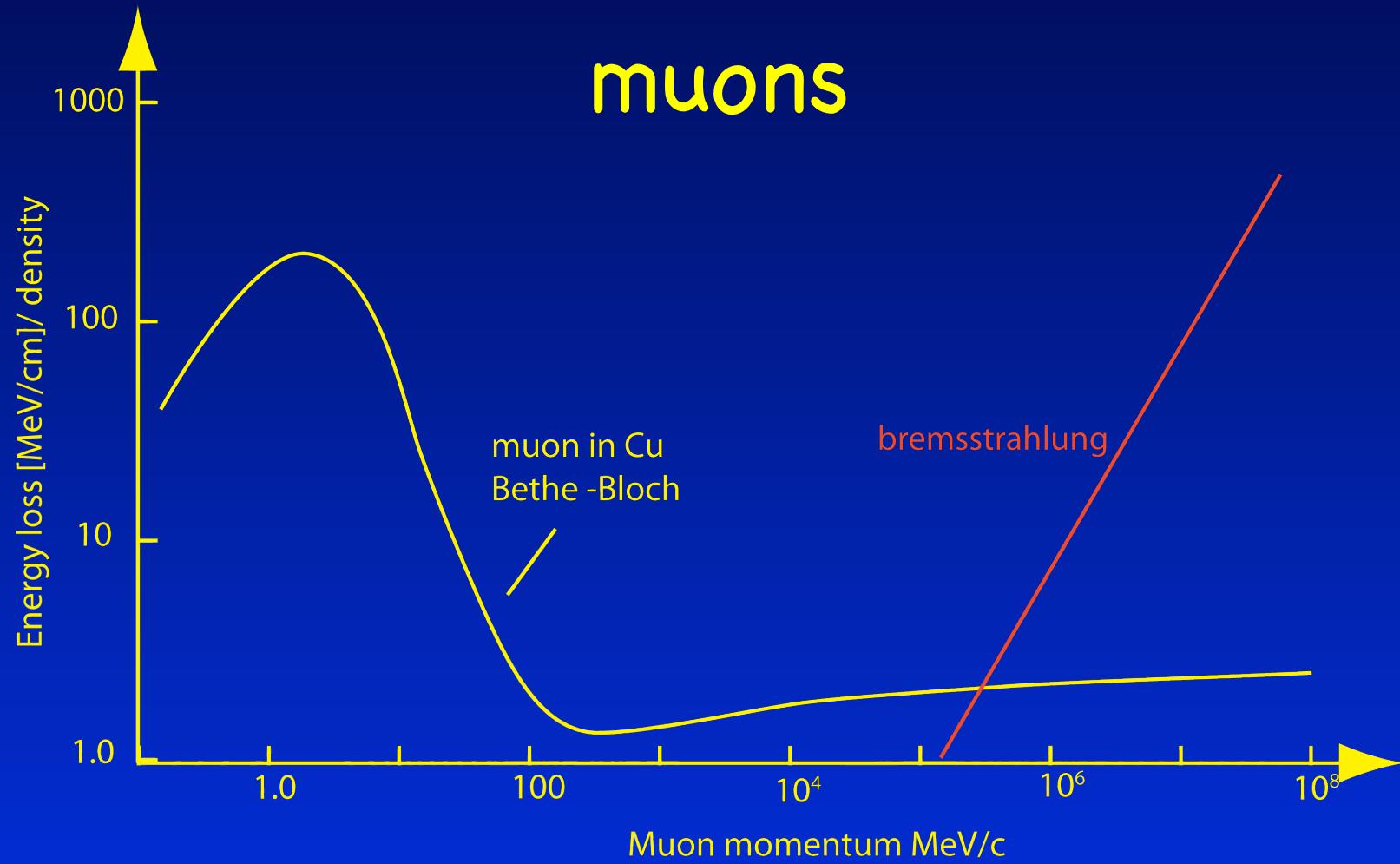


Will exceed energy loss by ionisation if  $E > E_c = \frac{800 \text{ MeV}}{(Z + 1.3)}$

For particles  $\neq$  electron, emission is suppressed by a factor  $(m_{\text{electron}}/M)^2$ ; significant only at very high energy

# Summary interactions of low energy charged particles (10 MeV in silicon)





# Interactions of energetic photons in matter

## X-rays and gamma rays

- photoelectric effect
- compton scattering
- pair creation

# Photoelectric effect

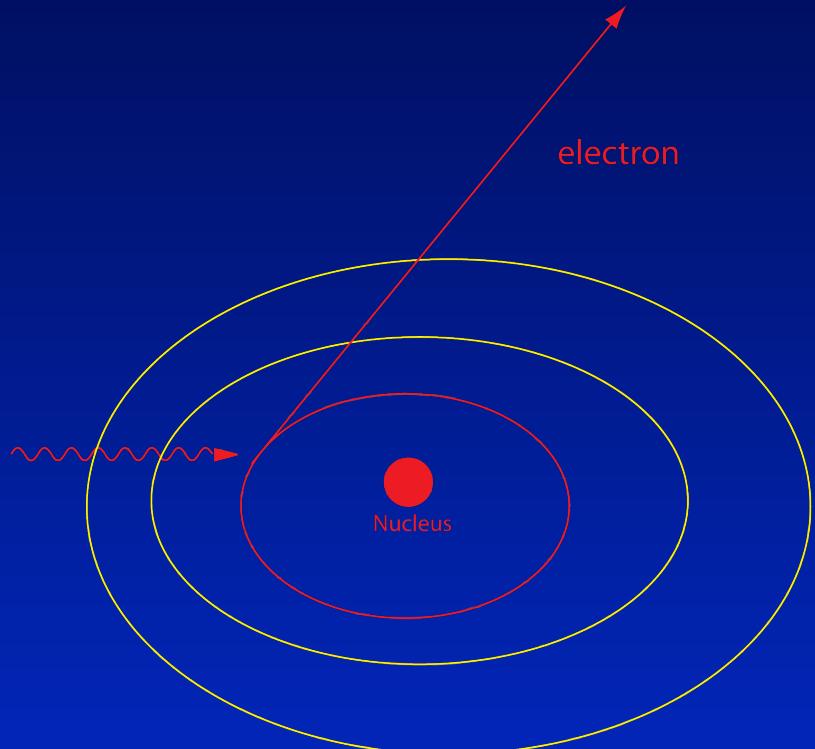
well above threshold >>

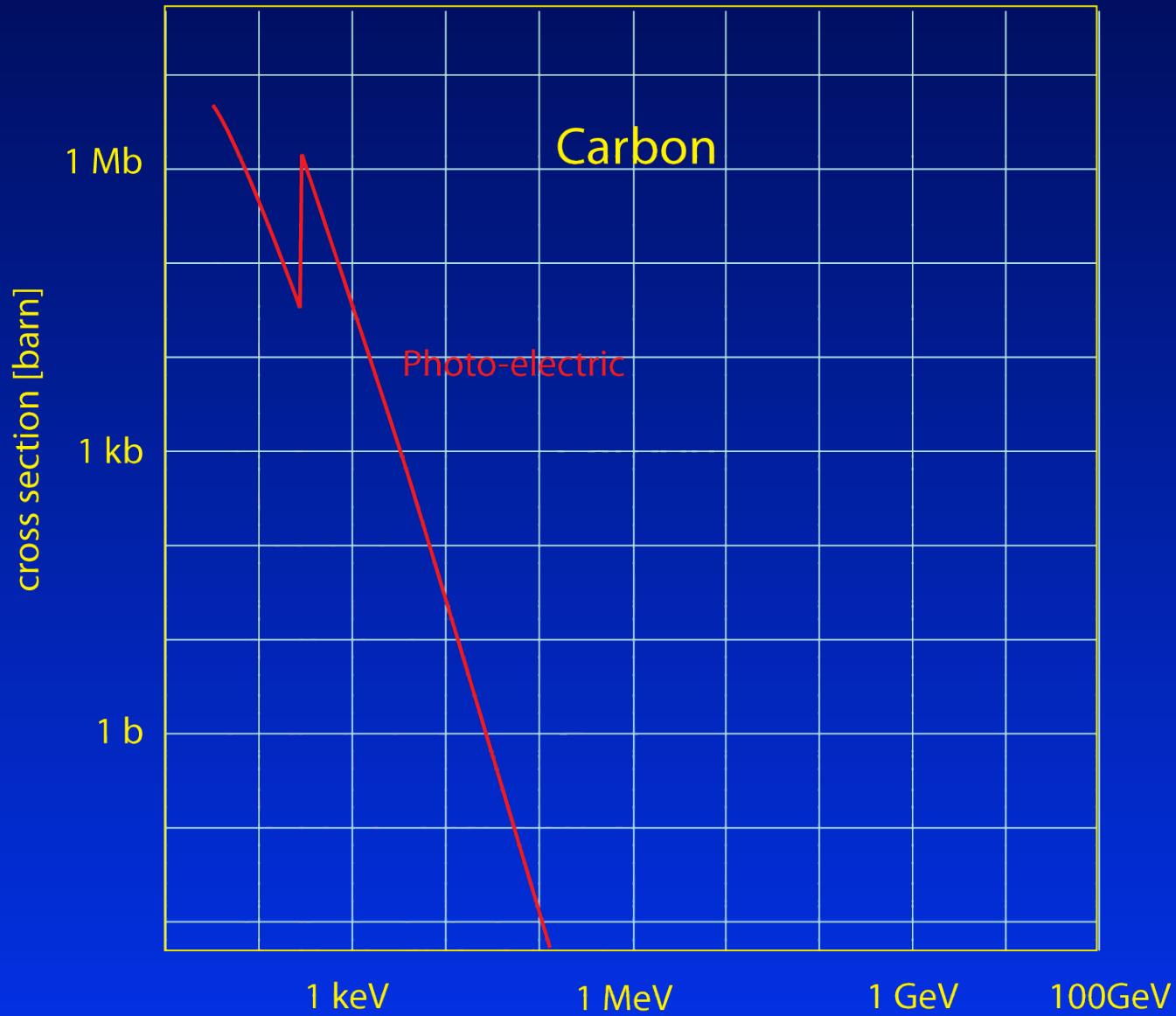
$$\sigma_{p.e.} = \frac{8\pi r_0^2}{3} 4\sqrt{2} \alpha^4 Z_{Nucl}^5 \left( \frac{m_e c^2}{E_\gamma} \right)^{7/2}$$

$$E_{kinetic} = \hbar\omega - E_{binding}$$

+ X-rays or Auger electron

80% of time on innermost electron





# Compton scattering (on free electron)

$$\hbar\omega' = \frac{\hbar\omega}{\left(1 + \frac{\hbar\omega}{m_e c^2} (1 + \cos\theta)\right)}$$

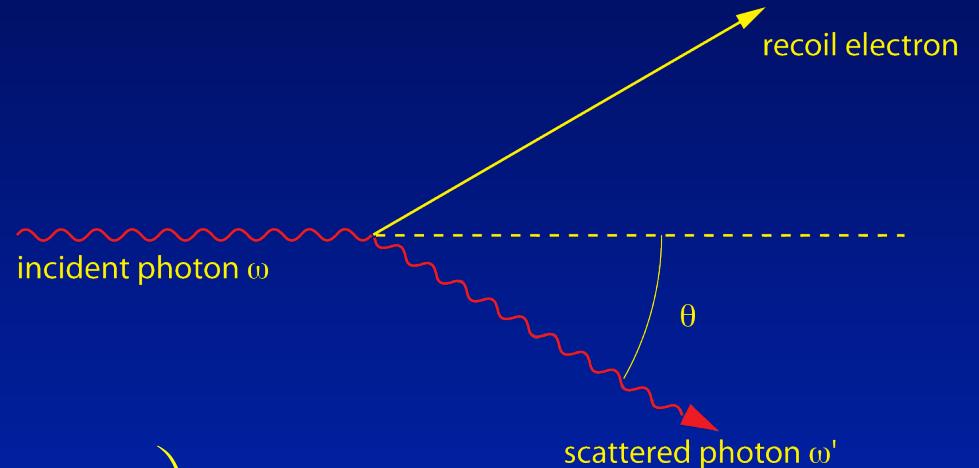
$$\frac{d\sigma}{d\Omega} = \frac{r_0^2}{2} \left( \frac{\hbar\omega'}{\hbar\omega} \right)^2 \left( \frac{\hbar\omega}{\hbar\omega'} + \frac{\hbar\omega'}{\hbar\omega} - \sin^2 \theta \right)$$

$$\sigma = \frac{8\pi}{3} r_0^2$$

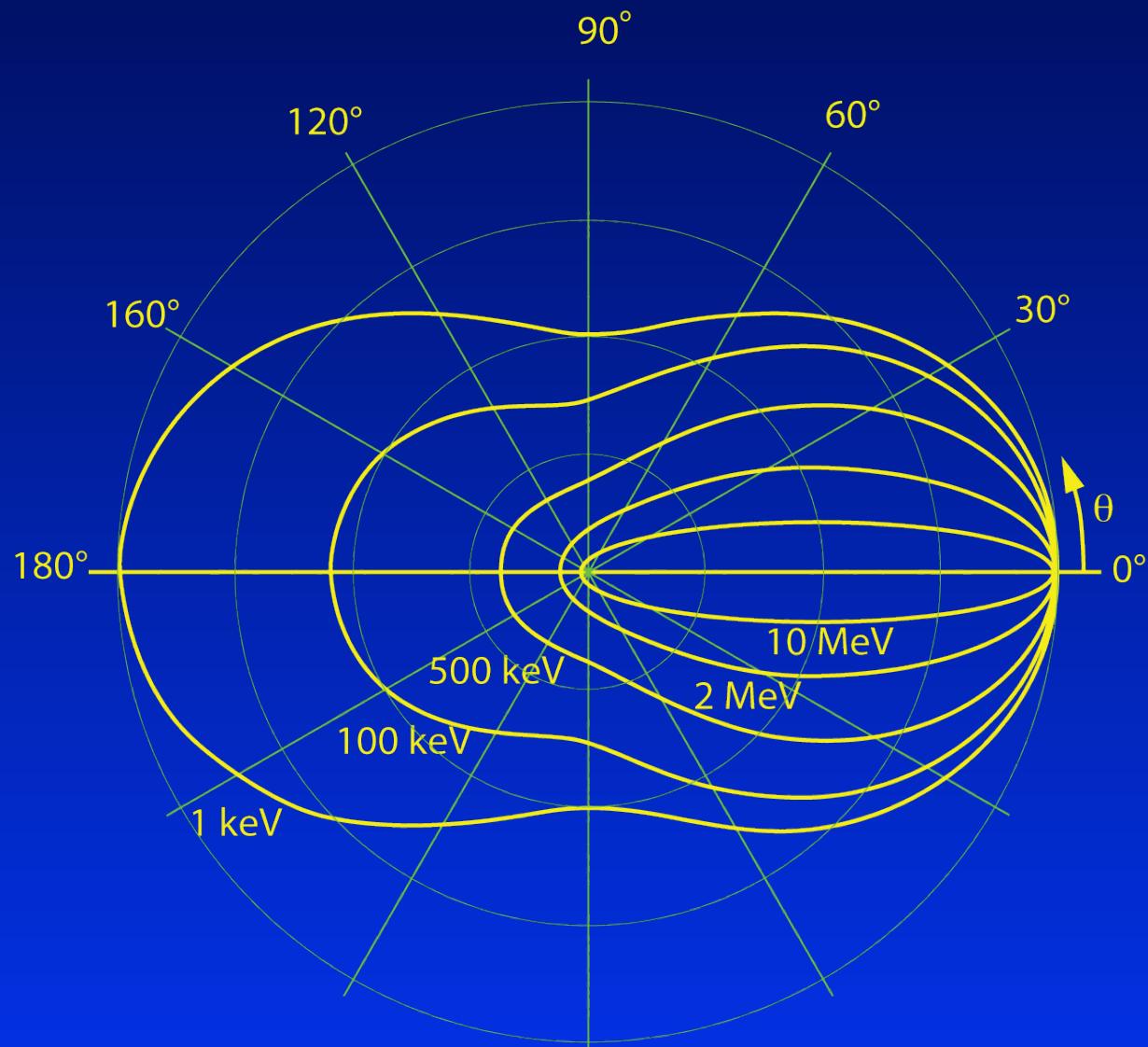
$$\sigma = r_0^2 \pi \frac{m_e c^2}{\hbar\omega} \left[ \ln\left(\frac{2\hbar\omega}{m_e c^2}\right) + \frac{1}{2} \right]$$

$\hbar\omega \ll m_e c^2$

$\hbar\omega \gg m_e c^2$



# Angular distribution Compton scattering



## Compton scattering (on electron bound in atom)

Usually on a weakly bound outer electron

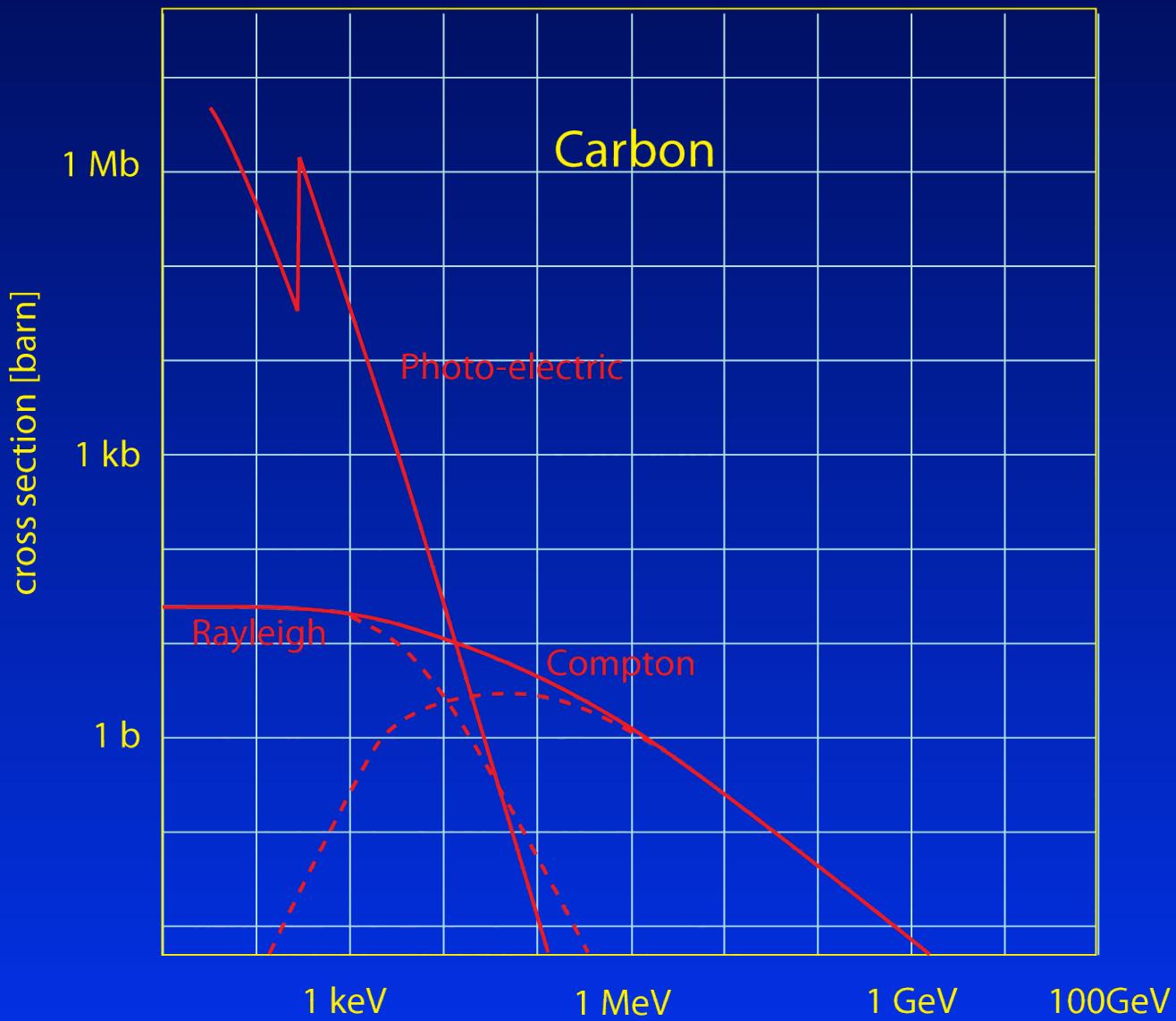
If electron remains bound in atom

=> coherent Compton, also called Rayleigh scattering

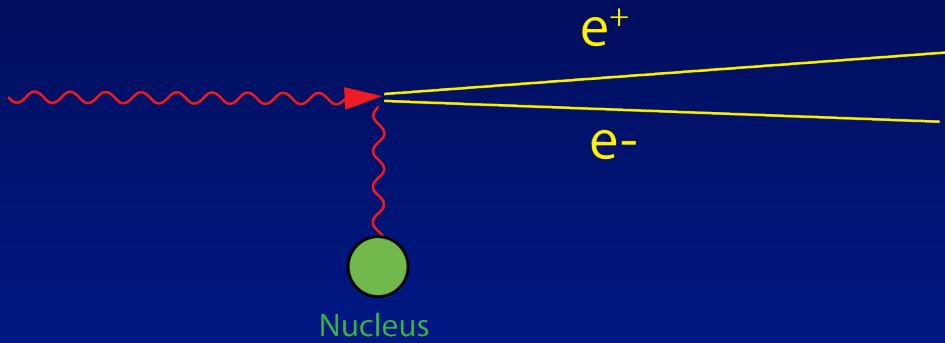
If atom ionised

=> incoherent Compton

Total cross section (coherent + incoherent), and angular distribution of scattered gamma still more or less the same as on free electron



# Pair creation



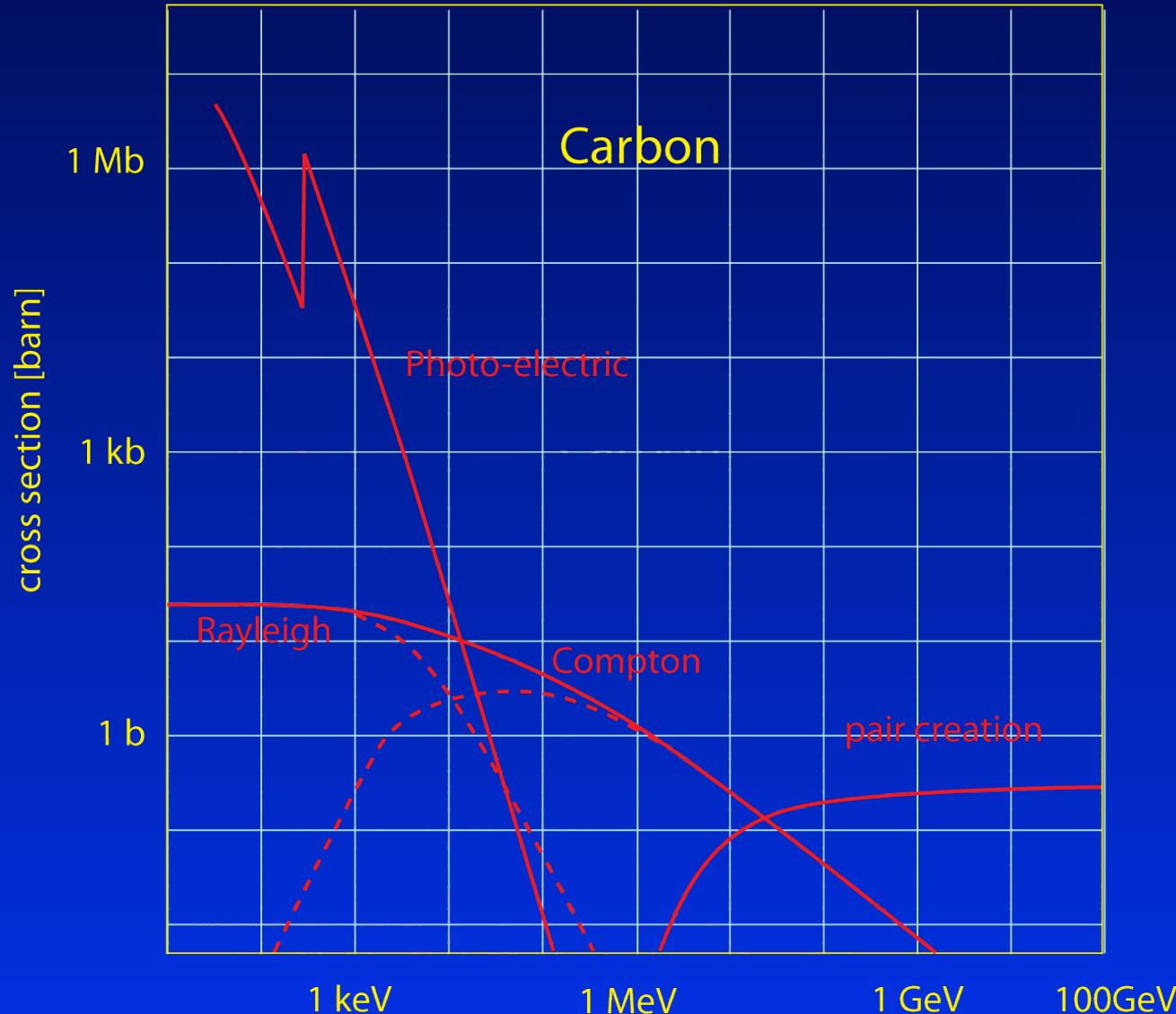
Threshold  $E > 2m_e c^2$

$$\frac{d\sigma}{dx} = \frac{1}{X_0} \frac{A_r}{\rho N_A} \left[ 1 - \frac{4}{3} k(1-k) \right] \quad ; \quad k = E_e / \hbar\omega \quad \Bigg| \text{ for } E >> 2m_e c^2$$

$$\sigma = \frac{7}{9} \frac{1}{X_0} \frac{A_r}{\rho N_A}$$

Probability pair creation     $dW = \frac{7}{9} \frac{dx}{X_0}$

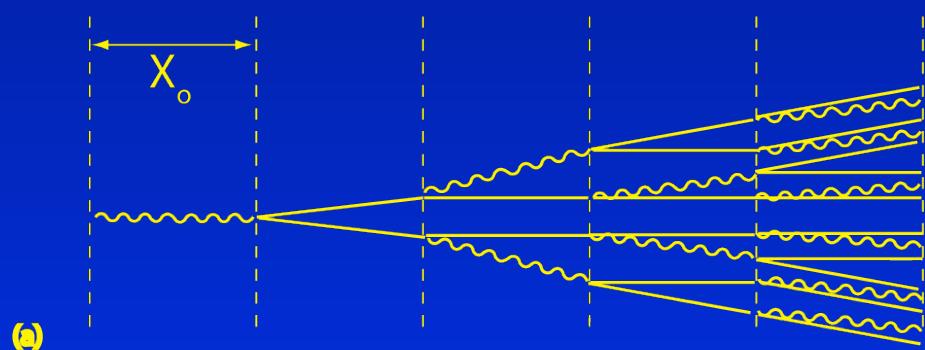
# Summary



# Interaction of a high energy gamma ray or electron in matter

Block of  $25 \approx$  radiation lengths

gamma  
→



$$X_0 = \frac{A_r \times 716.4 \text{ [cm}^{-2}\text{]}}{\rho Z_{nucl}(Z_{nucl} + 1) \ln\left(\frac{287}{\sqrt{Z_{nucl}}}\right)}$$

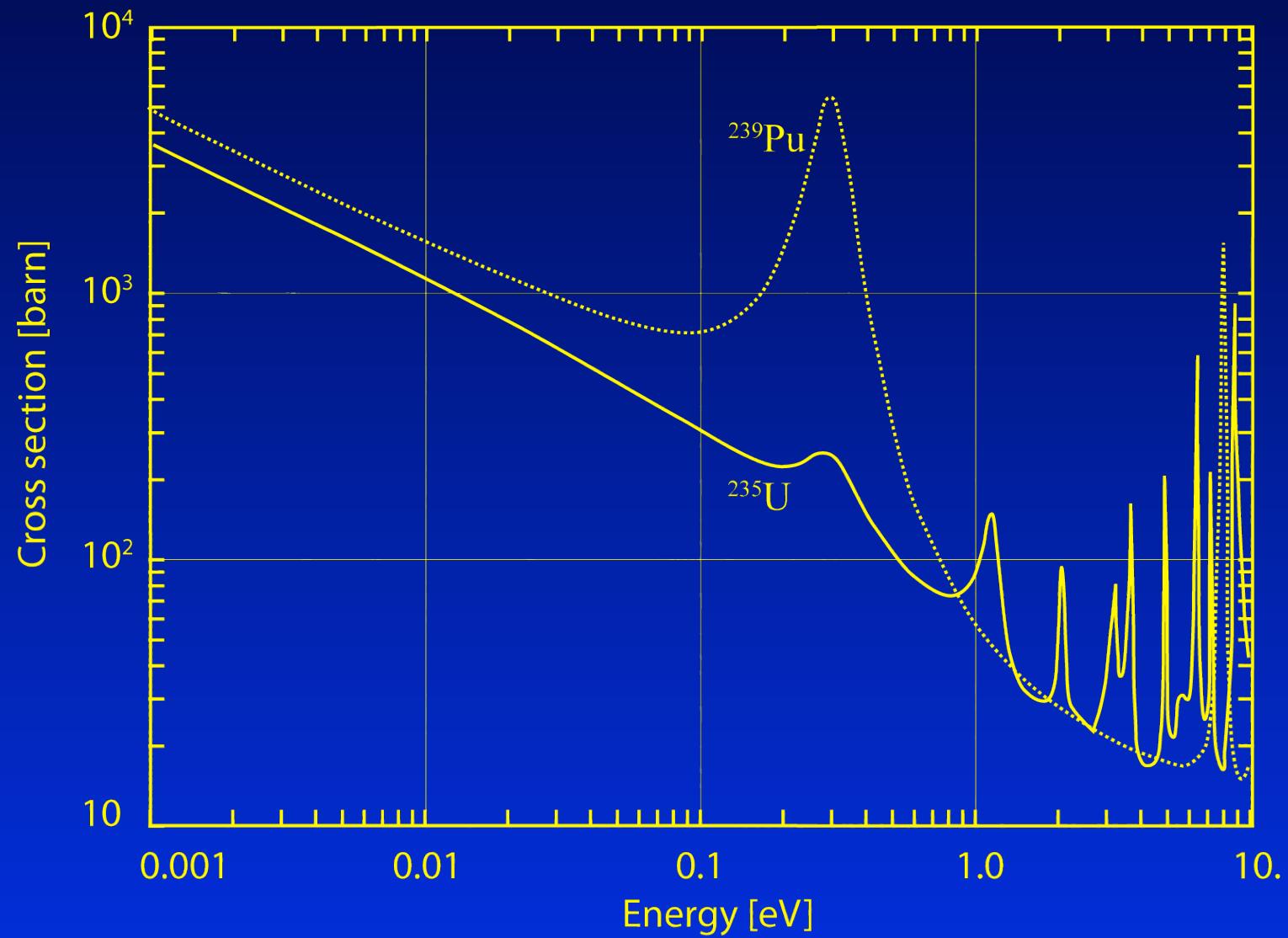
$$E_c \approx \frac{800 \text{ MeV}}{Z_{nucl} + 1.2}$$

# Nuclear interactions

## At MeV energy

Hadrons (bound states of quarks) undergo nuclear interactions: protons, neutrons etc. At MeV energies: strong resonance effects, cross section is variable and can be very large

- protons, alpha particles: will usually lose all energy by ionisation and just stop, occasionally undergo a nuclear interaction
- neutrons, will always end up making a nuclear interaction



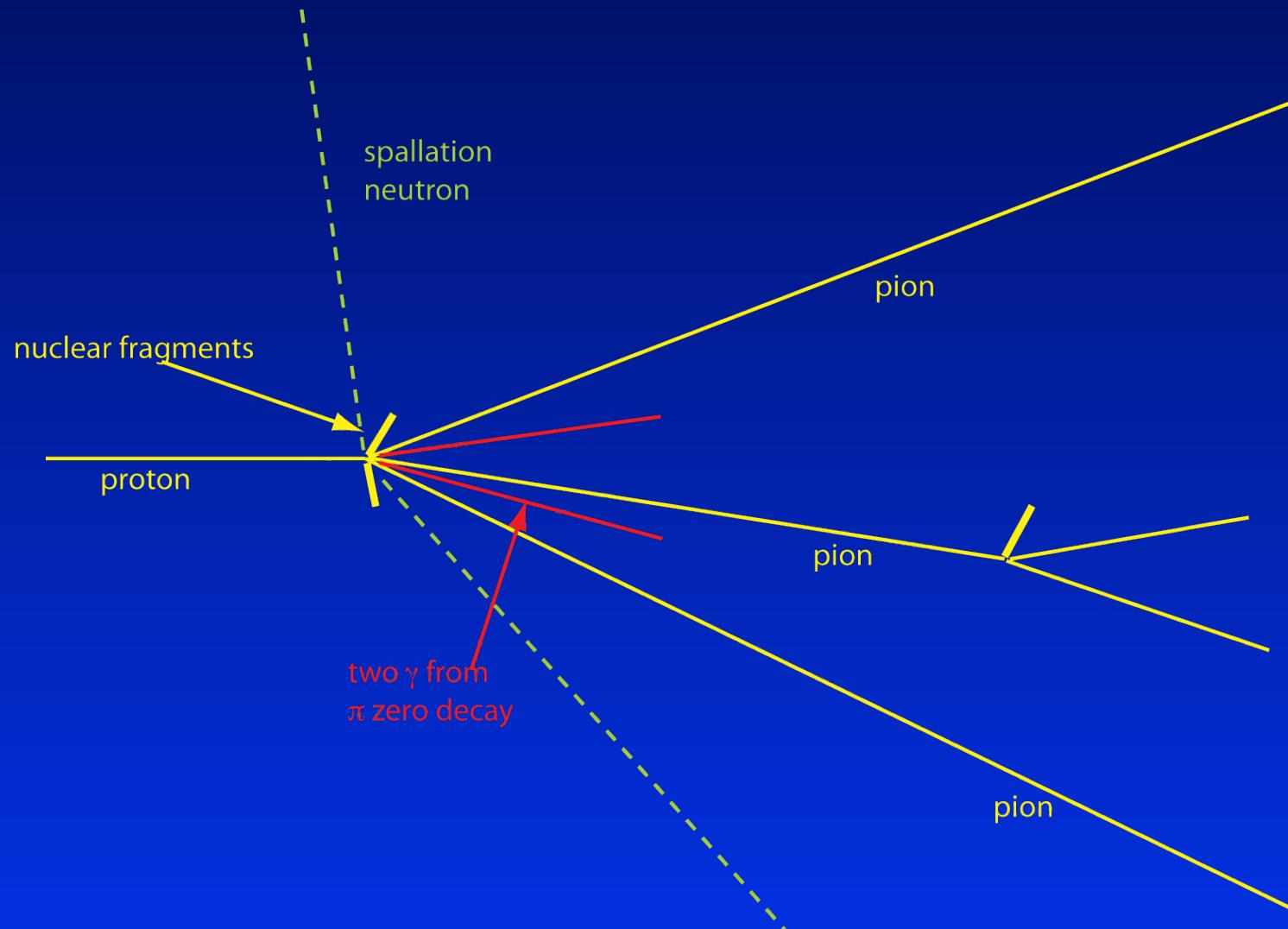
# Nuclear interactions At high energy $E \gg$ MeV

Constant cross section approximately the same for all hadrons

$$m.f.p. \approx \frac{A^{1/3}}{\rho} 35 g \text{ cm}^{-2}$$

m.f.p. hadrons in water  $\approx 60$  cm; in iron  $\approx 10$  cm

# Interaction of a high energy hadron in matter



Electrons, muons, gamma rays only rarely undergo nuclear interactions.

However, at very high energy ( $> 10^{20}$  eV) gamma rays, photonuclear interactions become dominant.

# Neutrino interactions

Neutrinos have no charge and are insensitive to nuclear interactions.

MeV neutrino      m.f.p.  $\approx 1$  light year on normal matter!

At high energy     $\sigma \propto E_\nu$

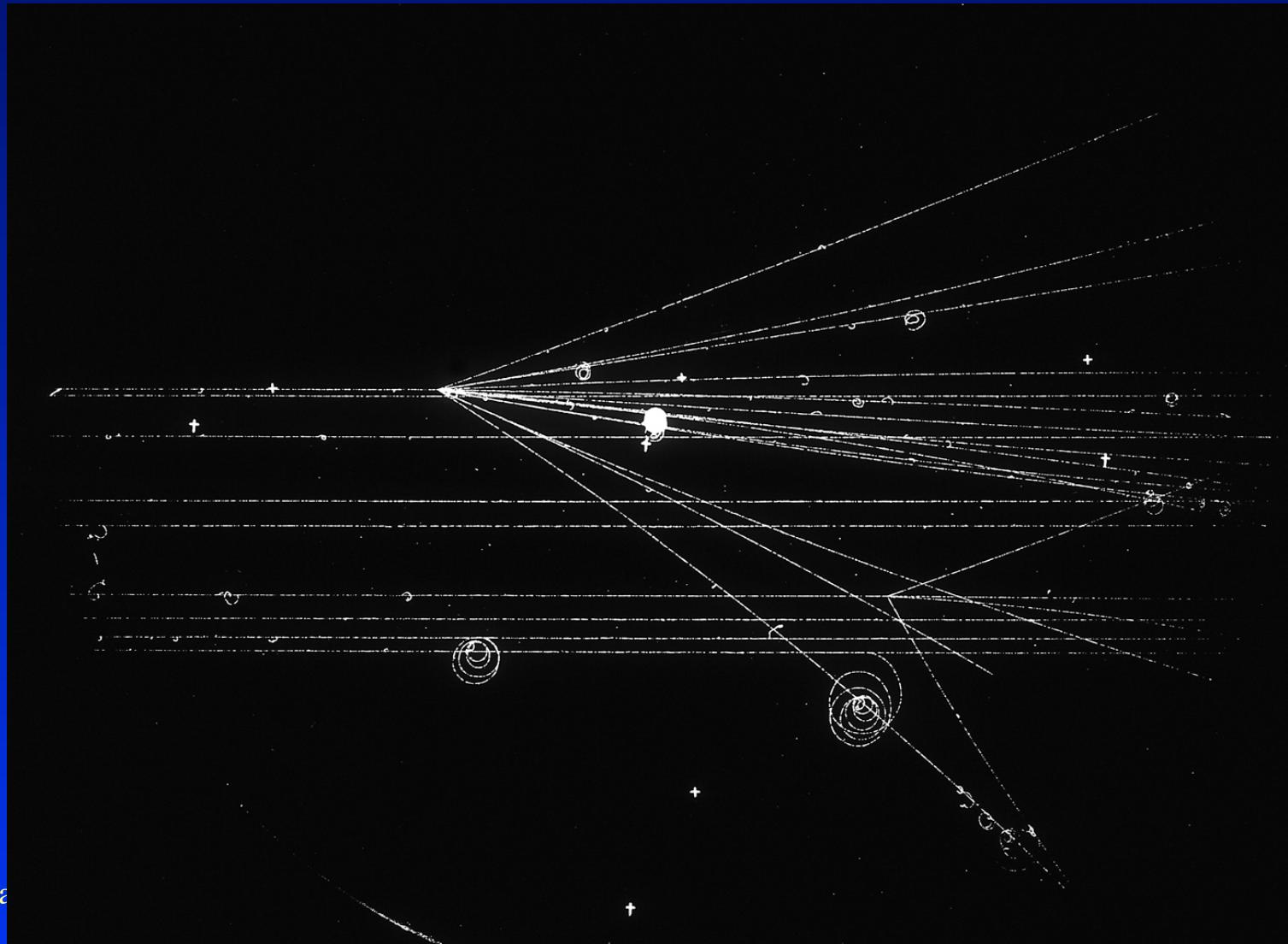
# Introduction to detectors and detector techniques

All detection of energetic subatomic particles is based in their electromagnetic interactions. Neutral particles must first interact and then one can detect the charged particles produced in the interaction.

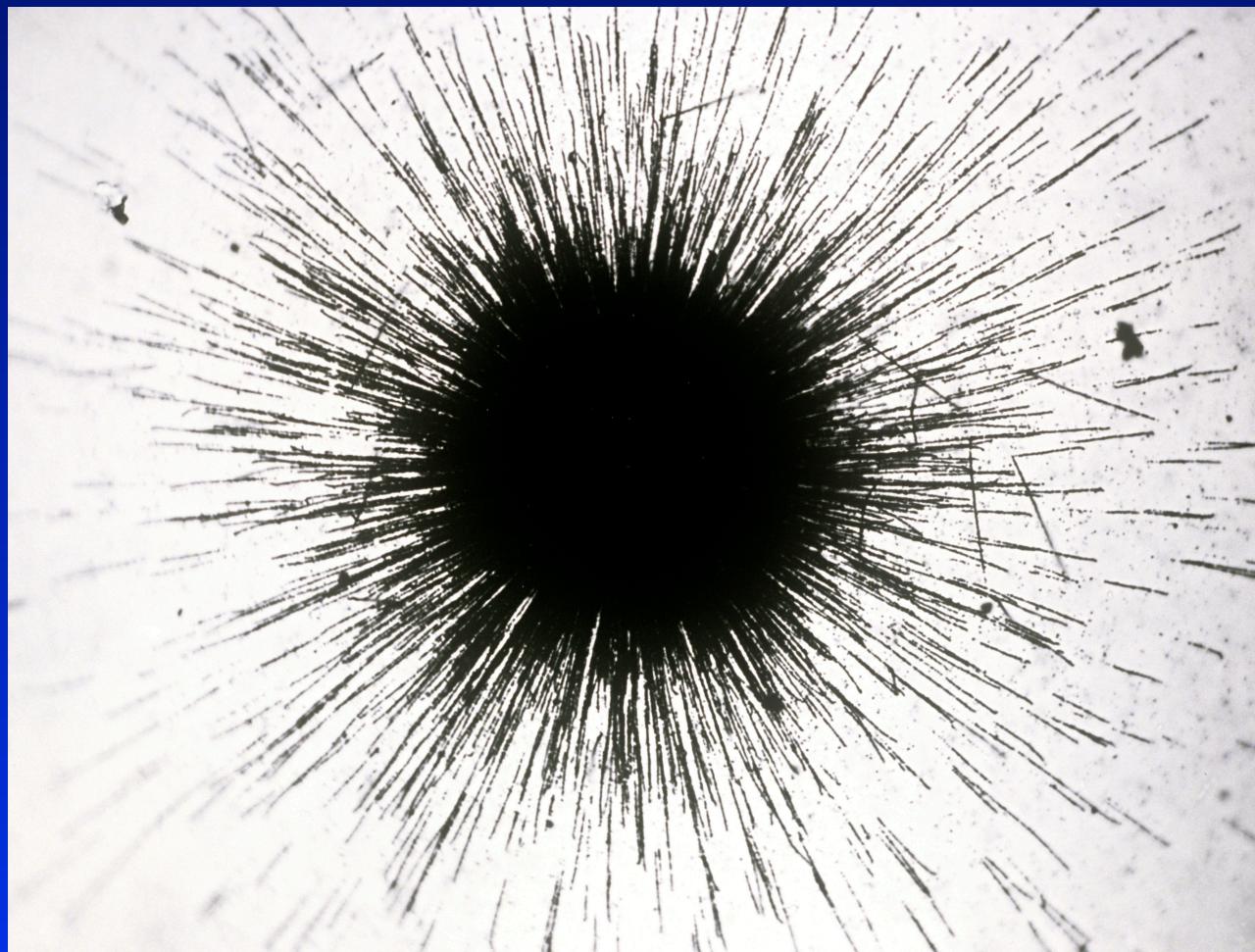
Most detectors are base on the ionisation produced by charges particles in matter. As a result, most detectors belong to one of the following categories

- detectors based on ionisation in gases
- detectors based on ionisation in semiconductors
- detectors based on scintillation

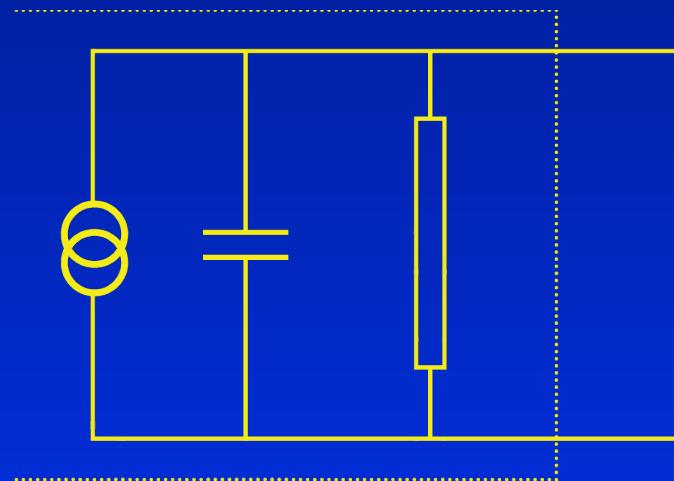
Years ago, many detection methods relied on making tracks visible, e.g. bubble chamber, spark chamber etc.



Most of these methods have fallen in disuse. One significant exception is the nuclear emulsion.



Today, nearly all detection is based on the fact that energetic subatomic particles induce some small electrical pulses . From the electronics point of view the detector is a current source, and it has a capacitance and a resistance.



# Current mode and pulse mode measurements

Simple instruments will only measure the average current in the detector. However, it is much better to observe the individual pulses. The pulses provide amplitude and time information that can be very useful.

- example amplitude:  $^3\text{He}$  tube: amplitude allows distinguishing neutrons from gamma induced pulses
- example timing: coincidence of two gamma rays characterises a positron annihilation

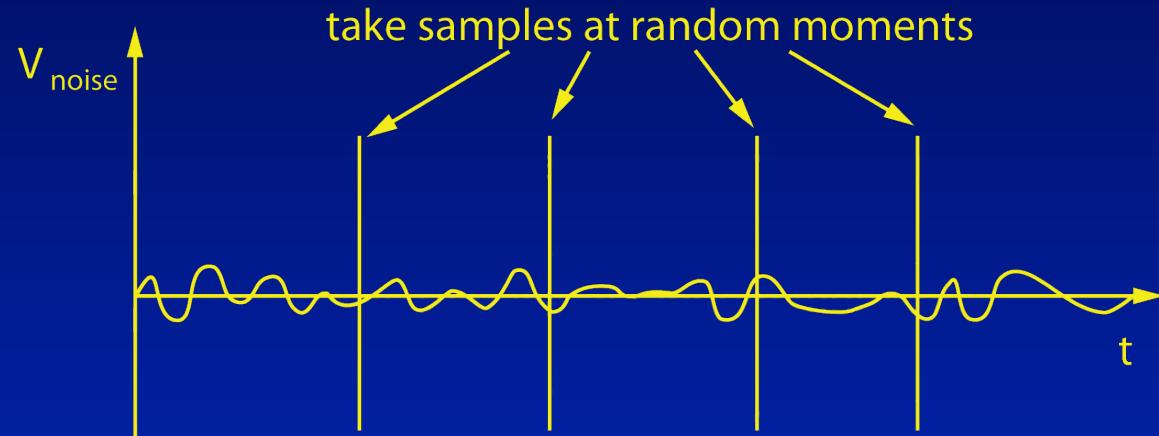
# Noise in detector systems

The signal from a detector is often very small. The signal can be difficult to see because of the electronic noise that is always present in any system. The two main types of electronic noise are

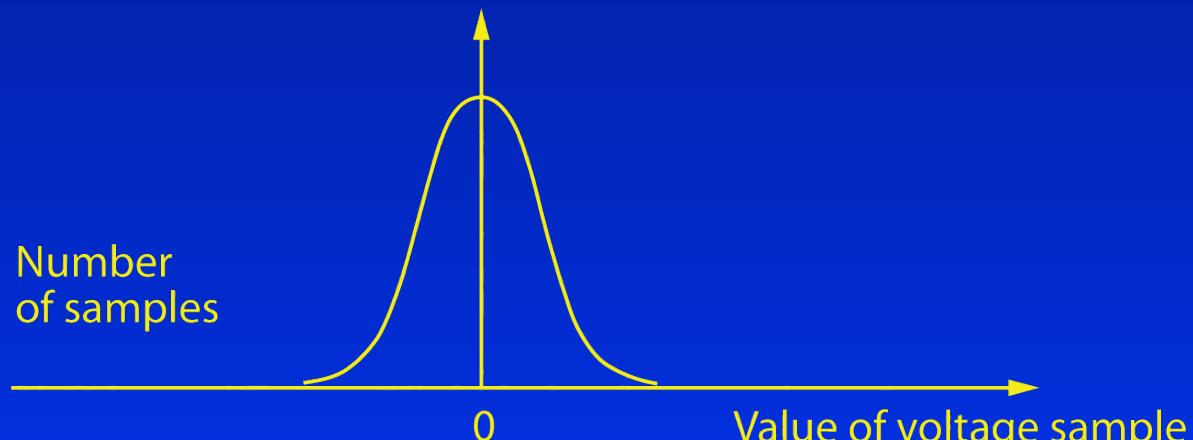
- pick-up noise, is avoidable in principle
- intrinsic noise, is unavoidable

In addition, there can be physical noise, i. e, real interactions of subatomic particles, but not the ones we want to see. This last type of noise is more often referred to as "background".

## The intrinsic noise has a smooth frequency spectrum



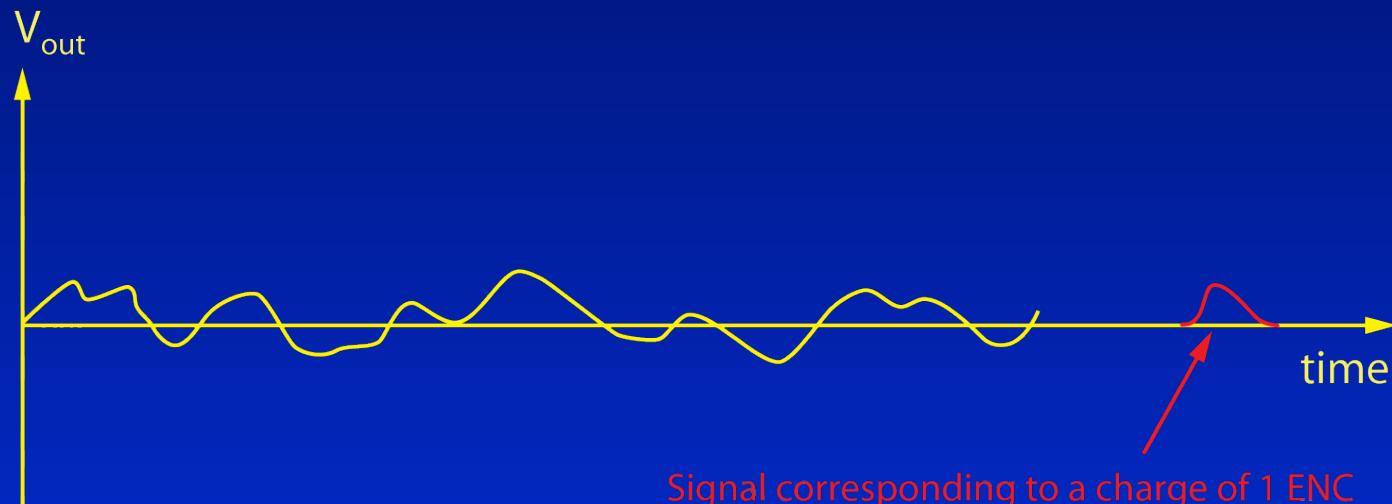
(a)



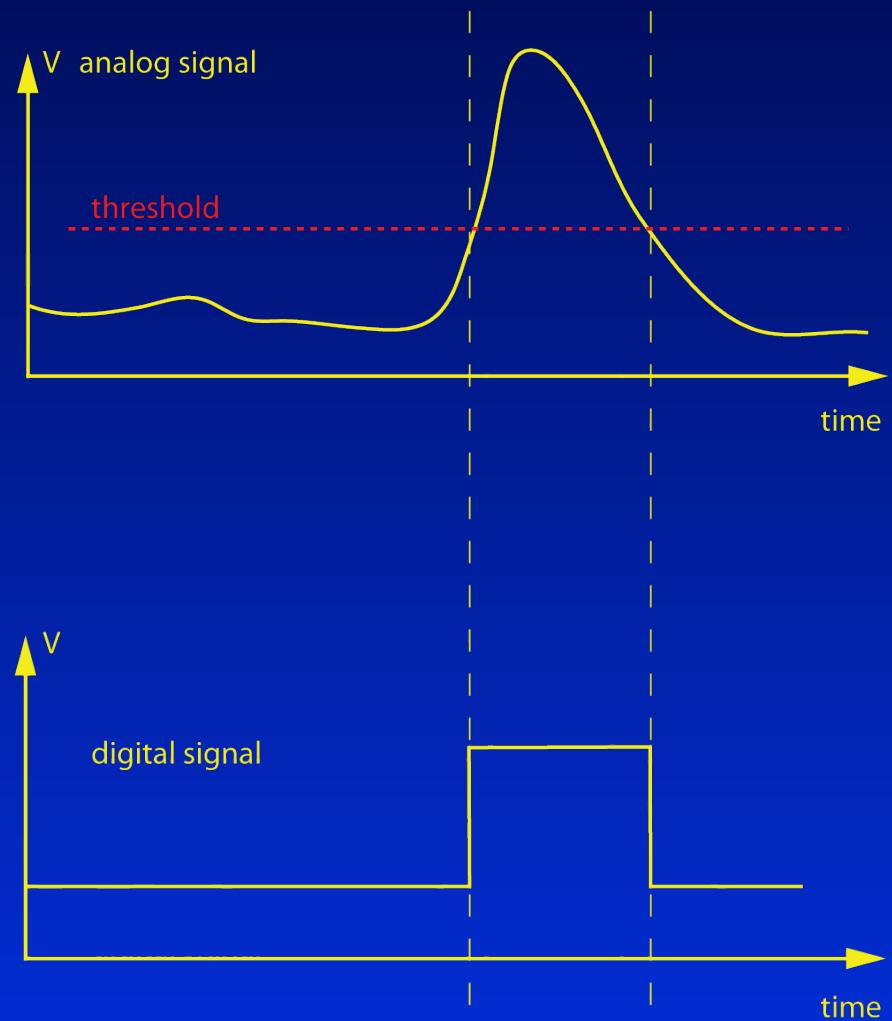
(b)

Definition of "equivalent noise charge" (ENC)

A short charge signal that gives a pulse such that the peak of the signal = one noise r.m.s.



The equivalent noise charge is usually expressed as a number of electrons



There is a signal if some threshold amplitude is crossed.

A signal of one equivalent noise charge is completely lost in the noise. A signal should be about 10 times the equivalent noise charge to be comfortable visible above the noise.

In a well designed system the pickup noise is small. The main sources of intrinsic noise are

- the shot noise caused by the dark current in the detector, the corresponding ENC is

$$\sqrt{2Q \tau I_{\text{dark current}}}$$

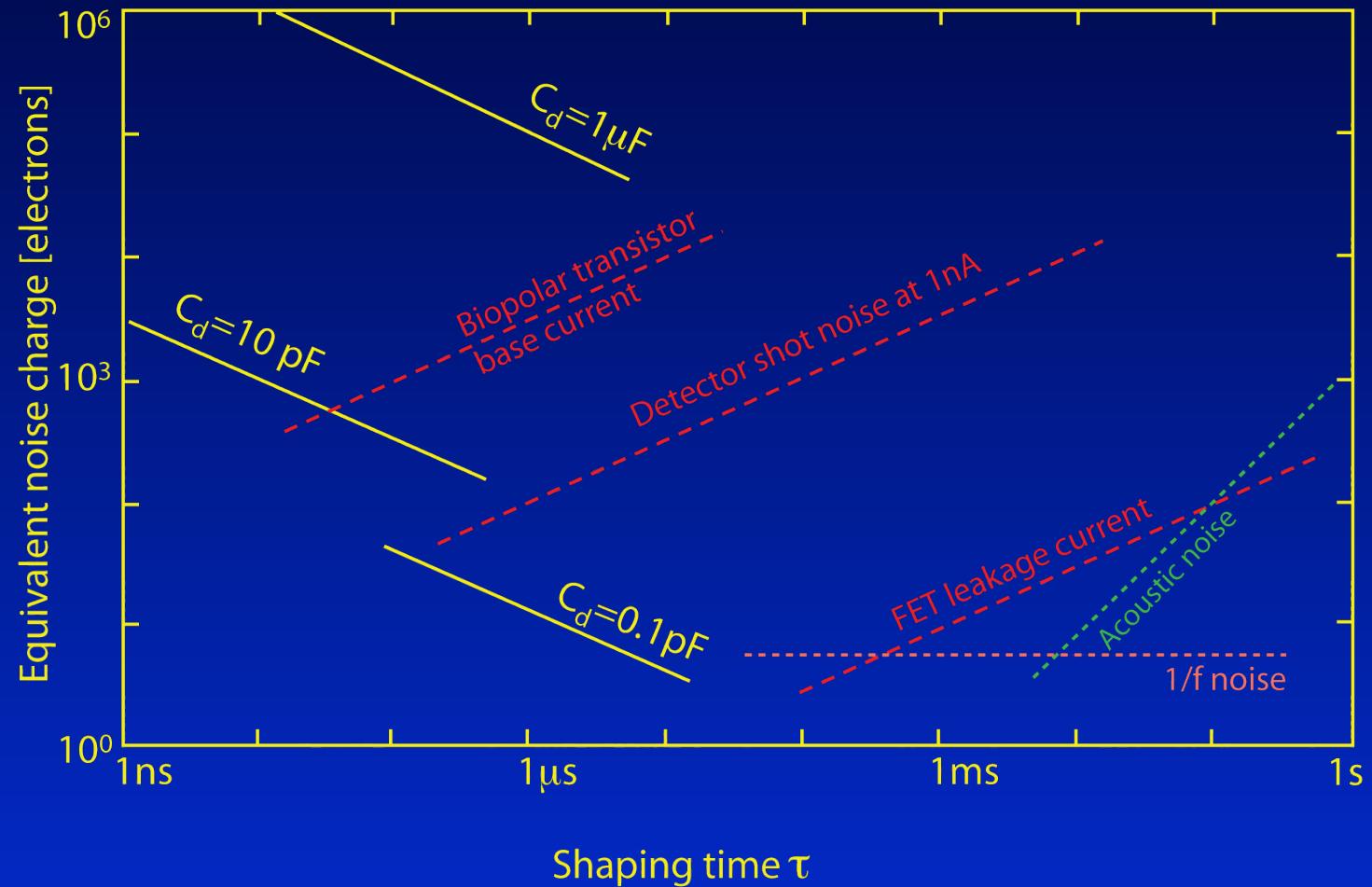
- the thermal noise of the first transistor, the corresponding ENC is given by

$$\sqrt{\frac{32}{3} kTC_d \frac{t_{\text{transit}}}{\tau}}$$

The shaping time  $\tau$  is related to the bandwidth (BW) of the amplifier. It determines the maximum count rate the system can handle

To have low noise the detector should have

- large internal resistance and therefore low leakage current
- low capacitance

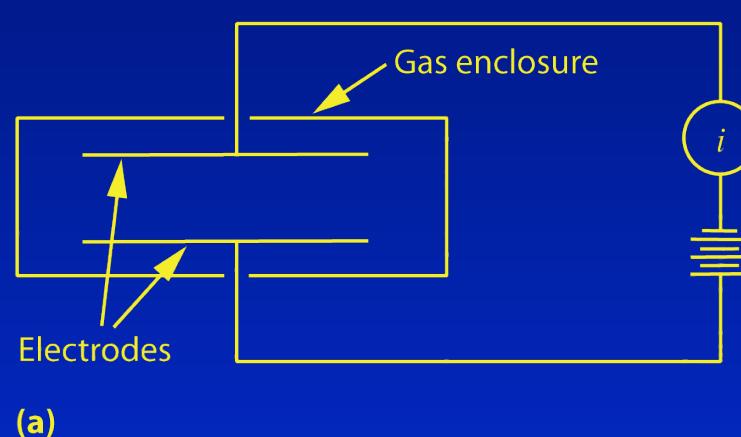


Overview of different contributions to the noise as a function of the pulse shaping time. Some other noise contributions are also shown.

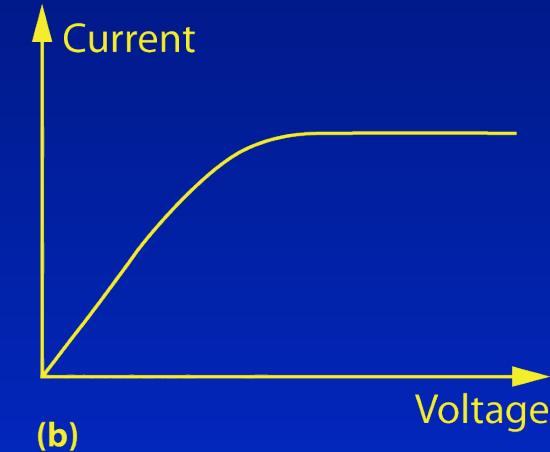
# Detectors based on ionisation in gases

# Basics of detectors based on ionisation in gasses

An electric field of a few 100 V is sufficient to collect ionisation in a gas



(a)



(b)

Let us first discuss the creation and movement of electrons and ions in a gas

**Table 4. 1: Energy loss characteristics in some commonly used gasses. Energy loss, the number of electron-ion pairs and the number of primary electrons is for charged particles at minimum ionisation.**

Gas	Ionisation potential [eV]	Mean energy /electron–ion pair ‘W’ [eV]	Energy loss [keV/cm]	Number of electron–ion pairs [cm <sup>-1</sup> ]	Number of primary electr.[cm <sup>-1</sup> ]
Ar	15.7	25.0	2.53	106	25
Xe	12.1	22	6.87	312	41
He	24.5	41.6	0.345	8.3	5
H <sub>2</sub>	15.6	36.4	0.32	8.8	5.2
N <sub>2</sub>	15.5	34.8	1.96	56.3	10
Air		33.8	2.02	59.8	
O <sub>2</sub>	12.5	30.2	2.26	74.8	22
CH <sub>4</sub>	12.6	30	1.61	54	37
C <sub>2</sub> H <sub>6</sub>	11.5	26	2.91	112	48
Isobutane/i-C <sub>4</sub> H <sub>10</sub>	10.6	26	5.67	220	90
CO <sub>2</sub>	13.8	34	3.35	100	35

Under the influence of an electric field the electrons and ions move

$$v = \mu E \quad , \quad \mu = \text{mobility}$$

For ions Ion-Ion collision cross section is independent of the kinetic energy of the ion

$$\rightarrow v_d = \frac{a}{2} \Delta t = \frac{eE}{2M} \frac{\lambda}{v_t} = \frac{e\lambda}{\sqrt{12kT M}} E$$

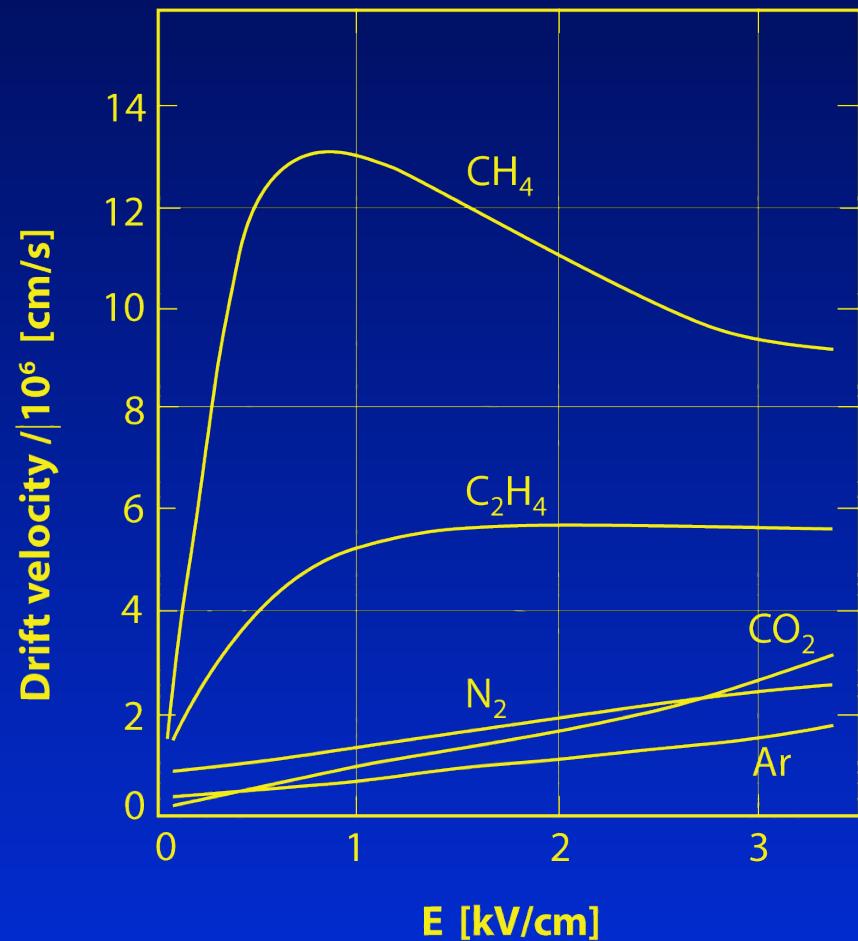
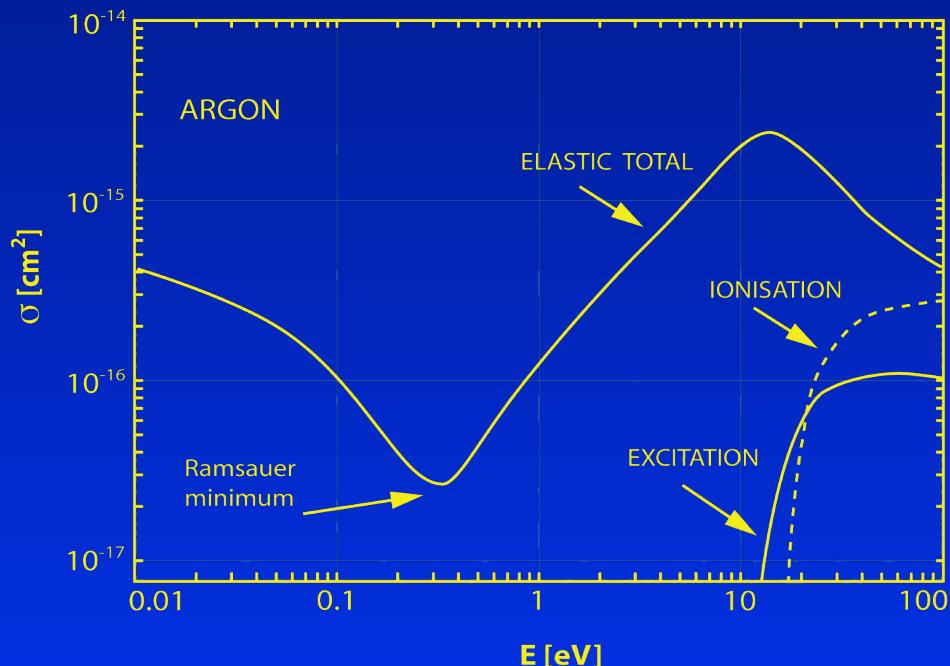
Mobility few [cm<sup>2</sup>/Vs]

at E=1000 V/cm >>>  $\rightarrow$  few 10<sup>3</sup> cm/s

For electrons drift velocity

$$v = \mu(E) E$$

Electron atom collision cross section is very dependent of the kinetic energy of the electron



A number of phenomena occur when electrons and ions drift in a gas in presence of an electric field

- if E sufficiently large, electron multiplication
- charge transfer:  $A^+ + B^0 \rightarrow A^0 + B^+$
- electron capture  $e^- + A^0 \rightarrow A^-$
- diffusion, longitudinal or transverse

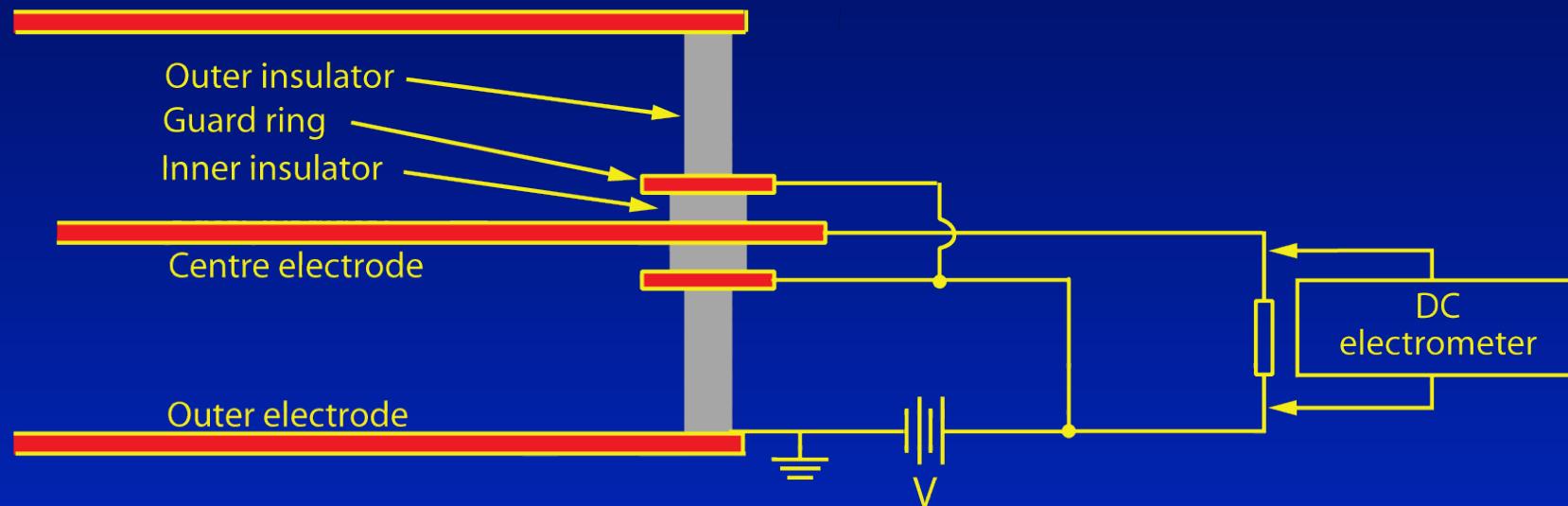
$$\sigma = \sqrt{(2D / v_d)l} \quad \mu = \frac{q}{kT} D$$

D=diffusion coefficient; l=drift distance

## What you should remember

- 1) For ions : ion-molecule collisions  $\sigma$  is independent of  $E_k$   
ion velocity is proportional to electric field E
- 2) For electrons
  - electron-molecule collisions  $\sigma$  is strongly dependent on  $E_k$
  - electron velocity is NOT proportional to electric field E
  - the mean free path (m.f.p.) of electrons  $\gg$  m.f.p. of ions
- 3) At fields of interest ( $\approx 1\text{kV/cm}$ )
  - ion velocity is typically few  $10^3 \text{ cm/s}$
  - electron velocity few  $10^6 \text{ cm/s}$
- 4) Charge multiplication sets in at much lower fields for electrons than for ions
- 5) Minimum ionising particles produce  $\approx 100$  electron-ion pairs /cm

# Ionisation chambers



Very small current, e.g. 1000 minimum ionising tracks/s  
current  $2 \times 10^{-14}$  A! Beware of leakage currents

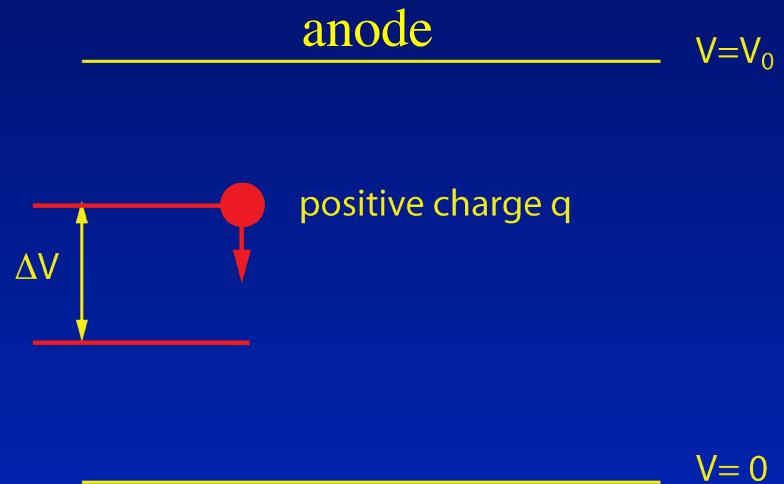
Consider only one ion. When does the signal appear on the anode? Use energy conservation!

$$d\left(\frac{1}{2}CV_0^2\right) = V_0CdV_0 = -q|\Delta V|$$

$$dV_0 = -\frac{q|\Delta V|}{CV_0}$$

$$\Delta Q = -\frac{q|\Delta V|}{V_0} = \frac{q \Delta V}{V_0}$$

$$i = \frac{dQ}{dt} = \frac{q}{V_0} \frac{dV}{dt} = \frac{q}{V_0} \frac{dV}{ds} \frac{ds}{dt}$$



Signal on cathode  
= minus signal on  
anode

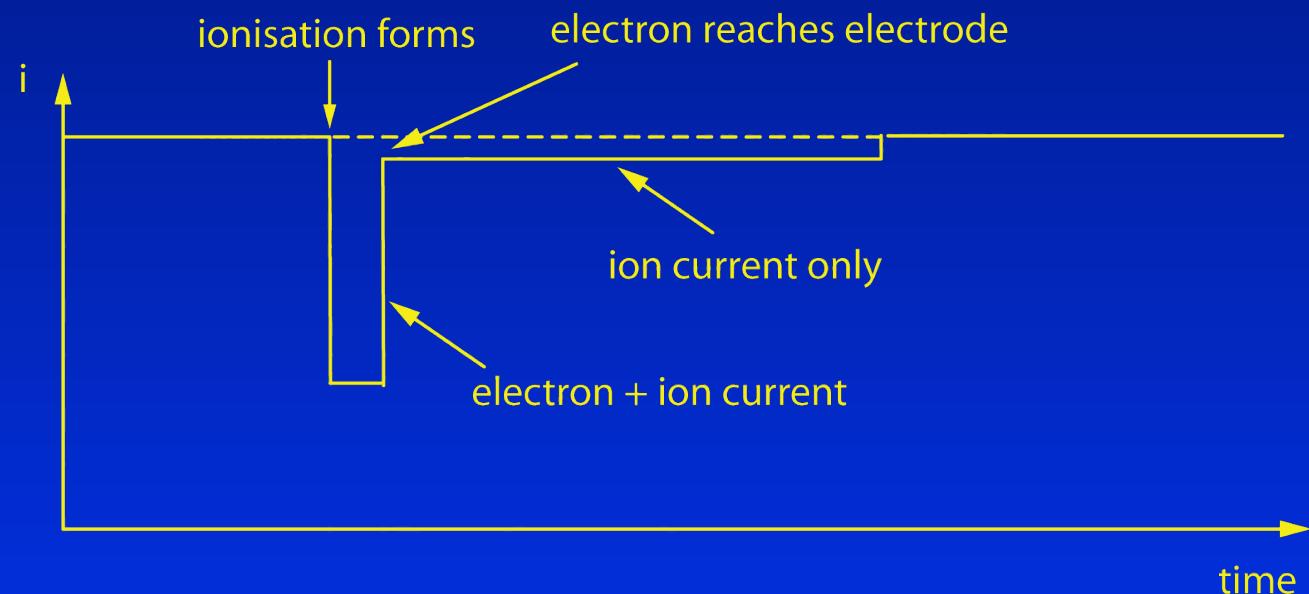
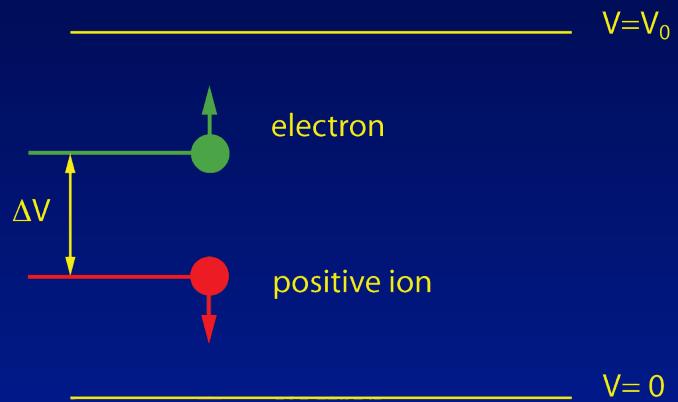
This is completely general, the motion of the charges induces an apparent current in the electrodes given by

$$i = \frac{q}{V_0} \frac{dV}{ds} \frac{ds}{dt}$$

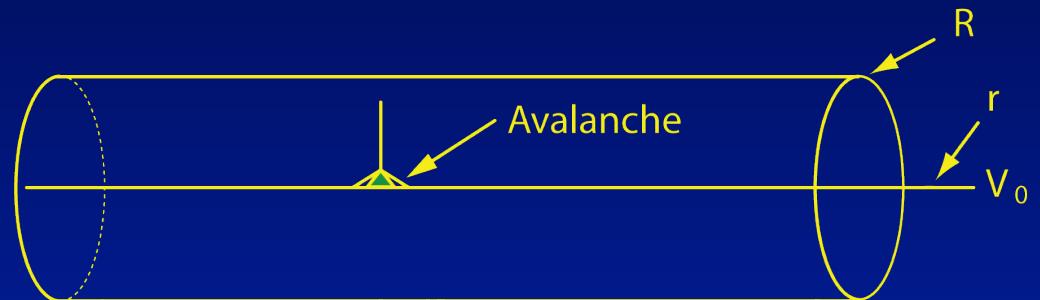
If more than 2 electrodes, same equation but the potential used in calculating  $dV/ds$  is the weighting potential (Cramer Rao theorem).

Weighting potential: all electrodes at  $V=0$ , except the electrode under consideration at  $V=V_0$

The signal of one electron ion pair. The signal is very small!



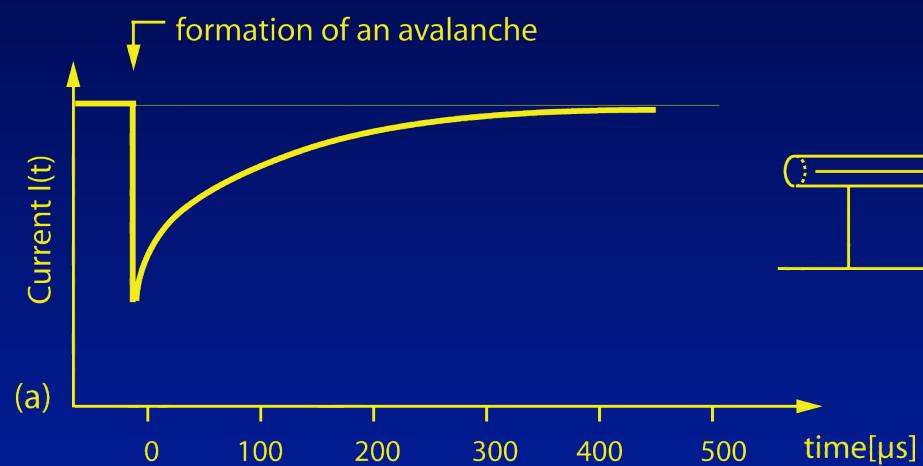
Electron multiplication allows getting a larger signal



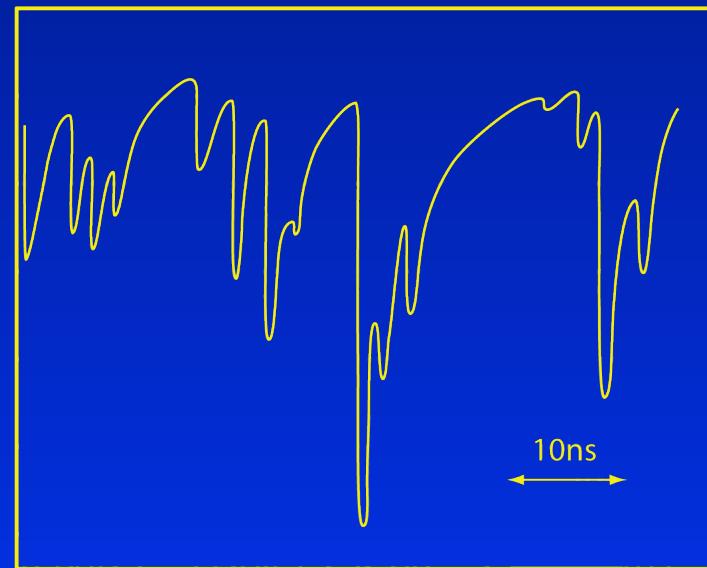
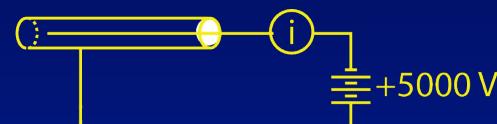
$$E(\rho) = \frac{V_0}{\rho \ln(R/r)}$$



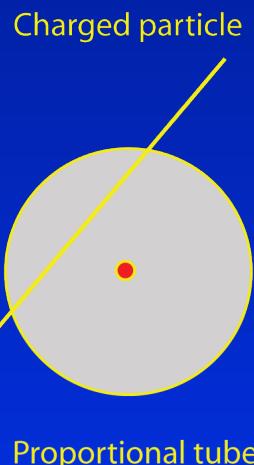
The signal comes almost entirely from the motion of the positive ions

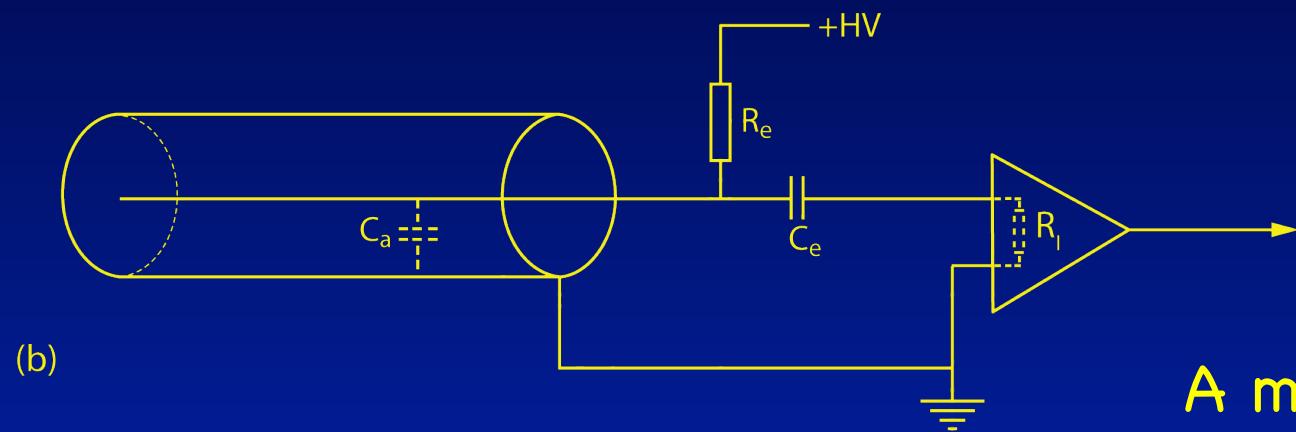


Signal from one electron

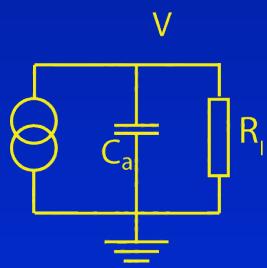
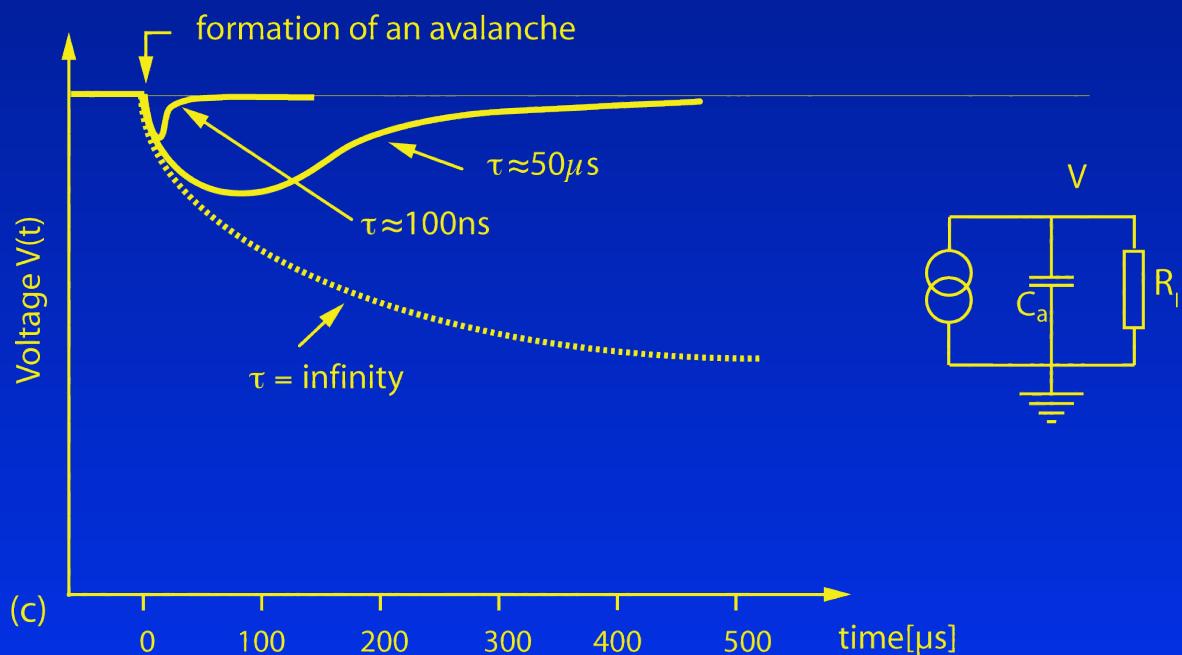


Signal from a track



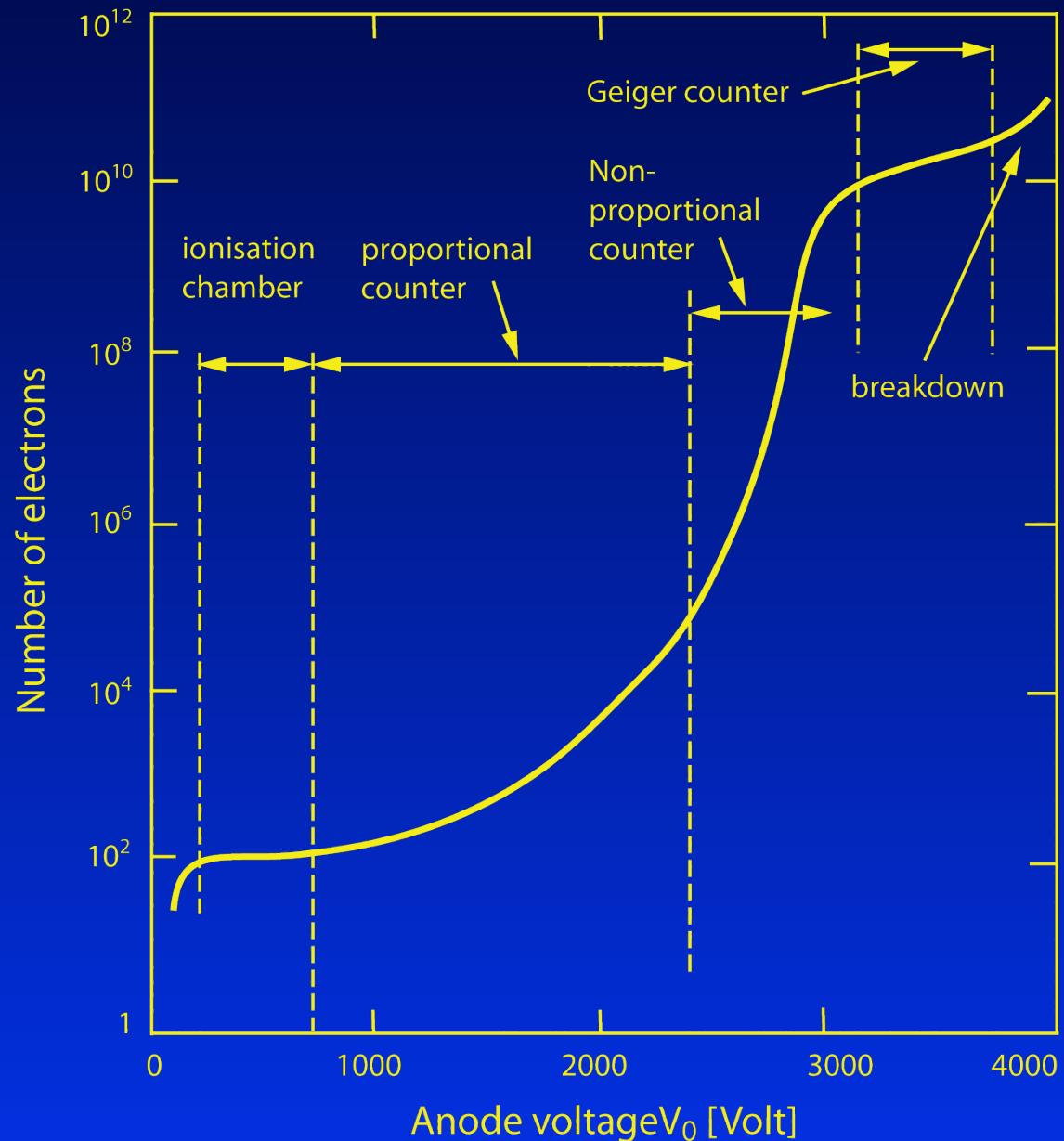


A more realistic  
readout electronics

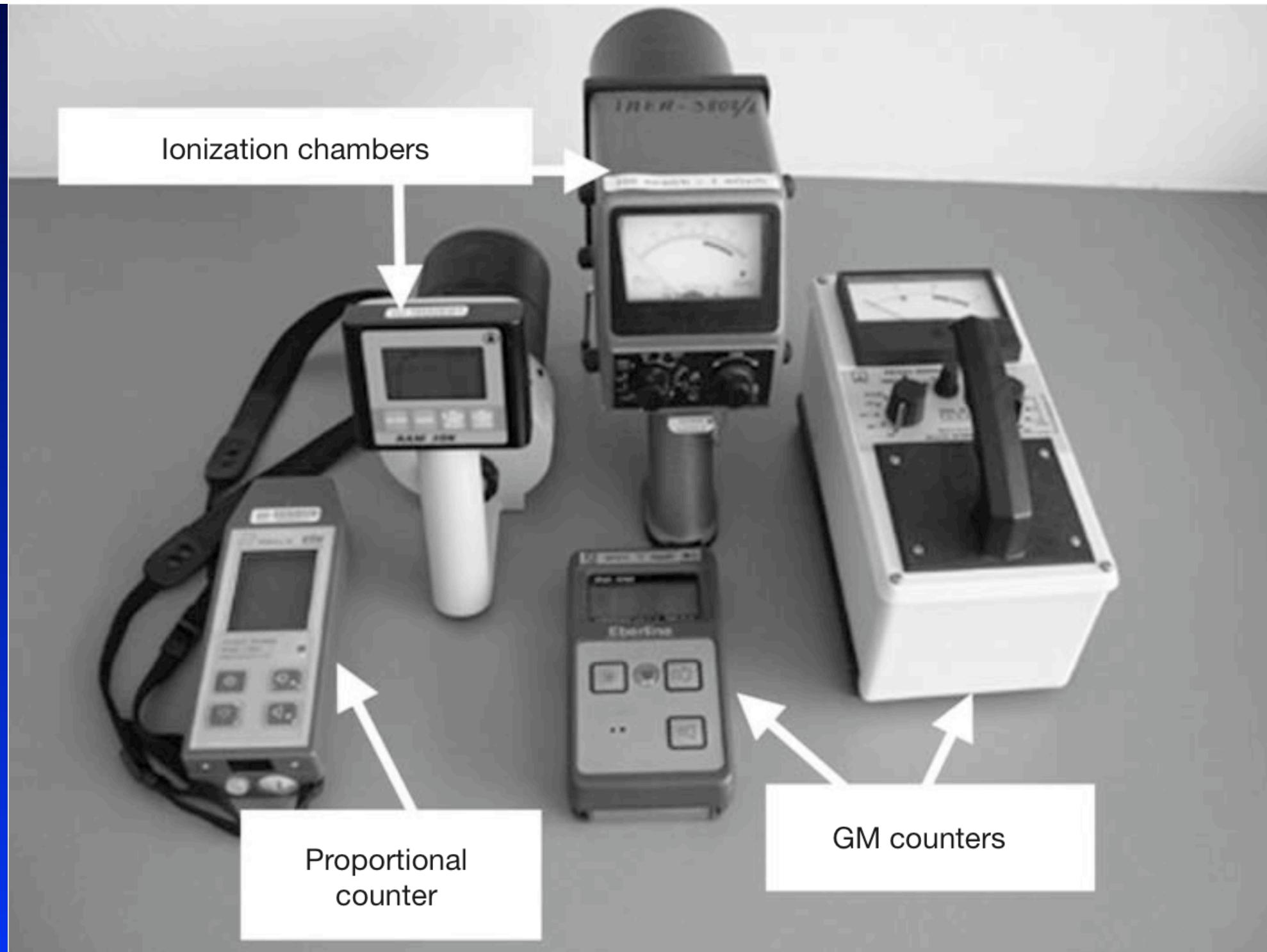


## Gas mixtures

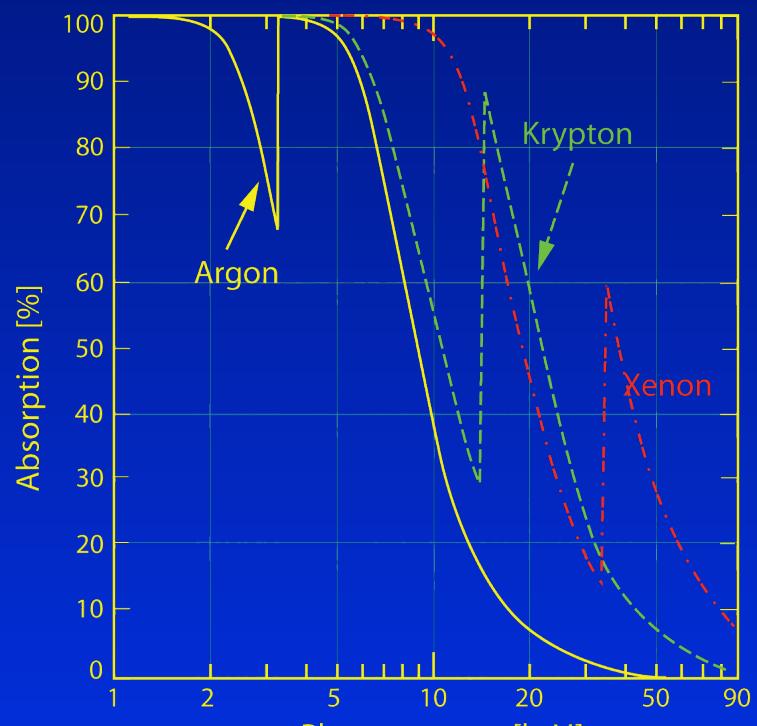
- the gas should not contain any electronegative component, no  $O_2$ , no air!
- noble gases ( e.g. Argon) are fine
- however, pure noble gas > problem
- need quenching gas, typically 10% of some polyatomic gas, e.g.  $CH_4$ , i- $C_4H_{10}$



# Instruments based on ionisation in gases



# Proportional tube as X-ray detector



absorption in 5 cm (no window!)

- detection efficiency

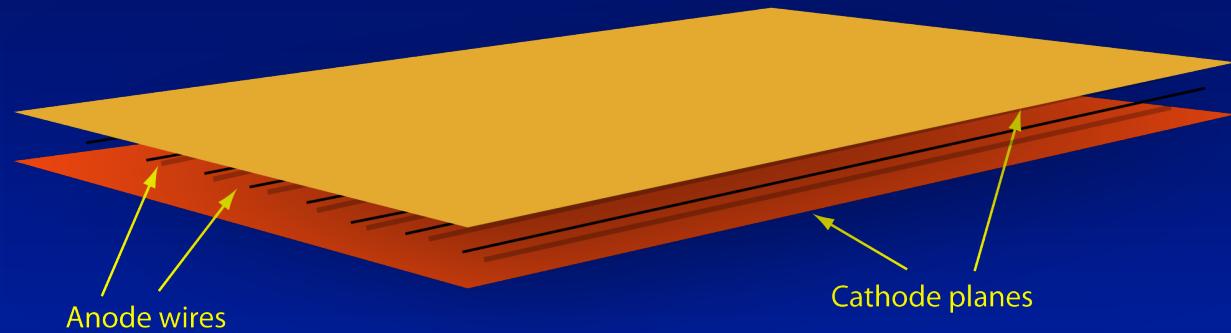
- energy resolution

$$\frac{FWHM[E]}{E} = 2.35 \sqrt{\frac{(F_{fano} + b) W}{E}}$$

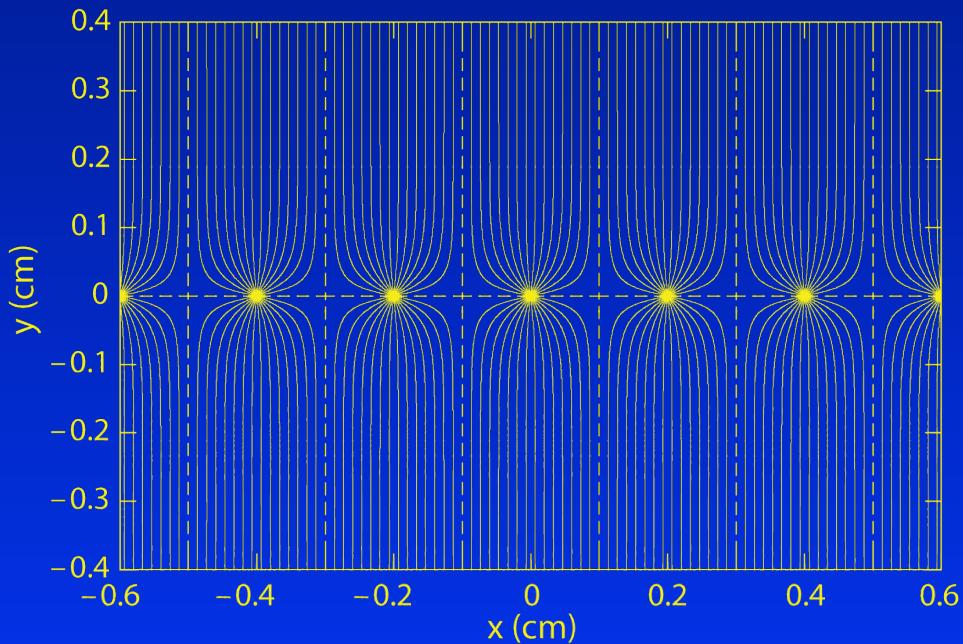
Fano factor  $F_{fano}=0.05-0.20$

Gain fluct.  $b = \frac{\sigma^2\{g\}}{\langle g \rangle^2} \approx 0.4 - 0.7$

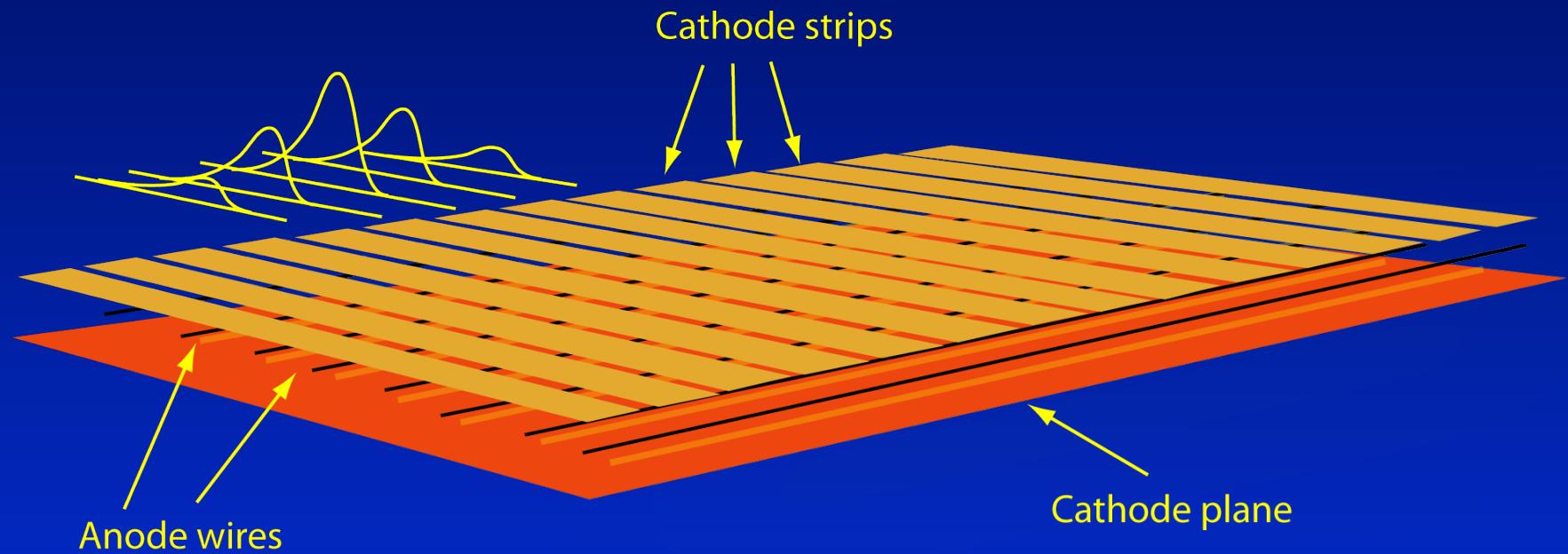
The main application of a gas chamber is for particle tracking  
multi-wire proportional chamber



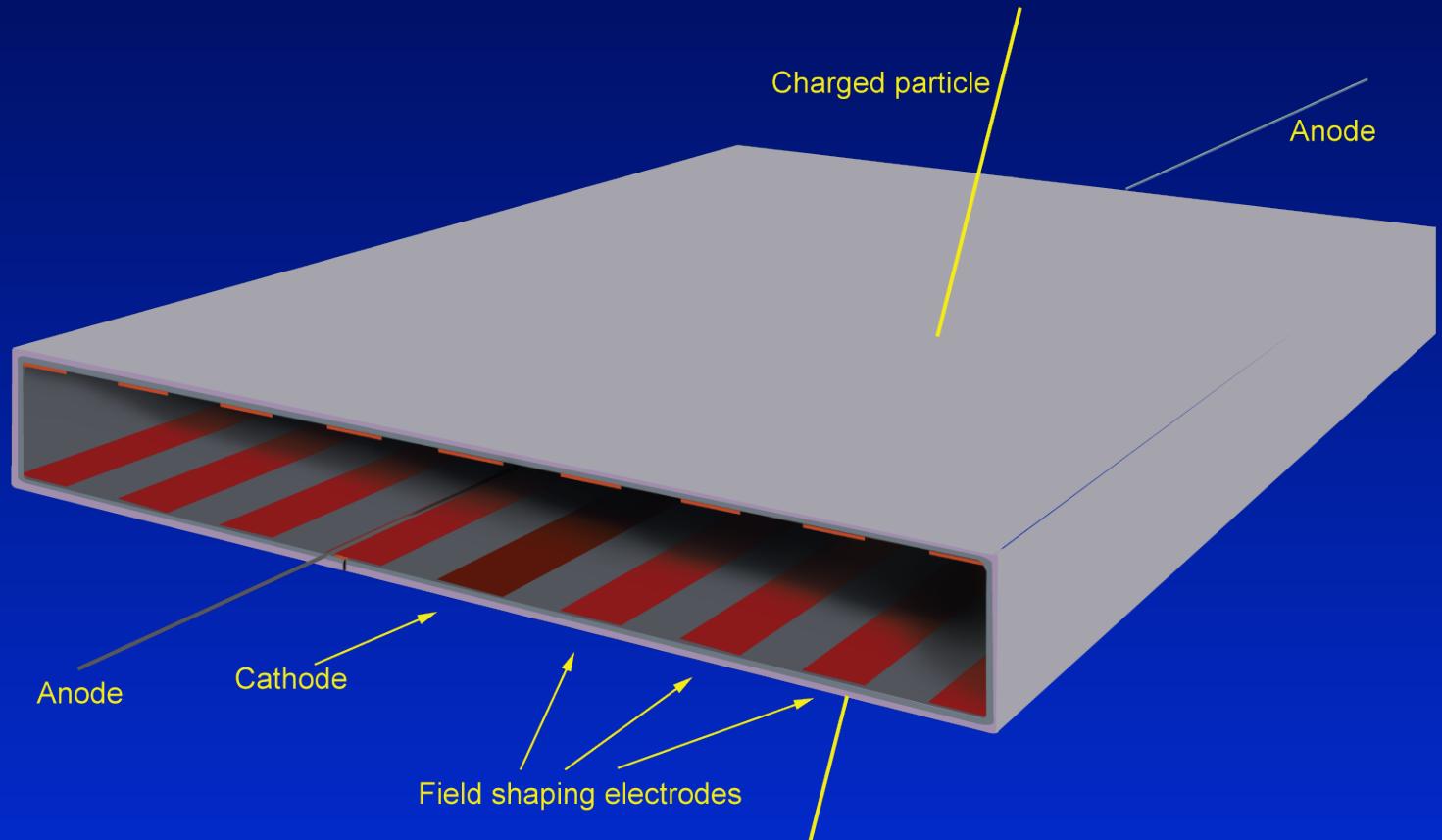
Wire spacing  $\approx 2\text{mm}$   
limited by electrostatic  
repulsion  
Spatial resolution  
spacing/ $\sqrt{12}$  mm



Cathode strip readout allows a better spatial resolution limited by range of delta electrons in gas r.m.s. $\approx$  50  $\mu\text{m}$



# Drift chambers



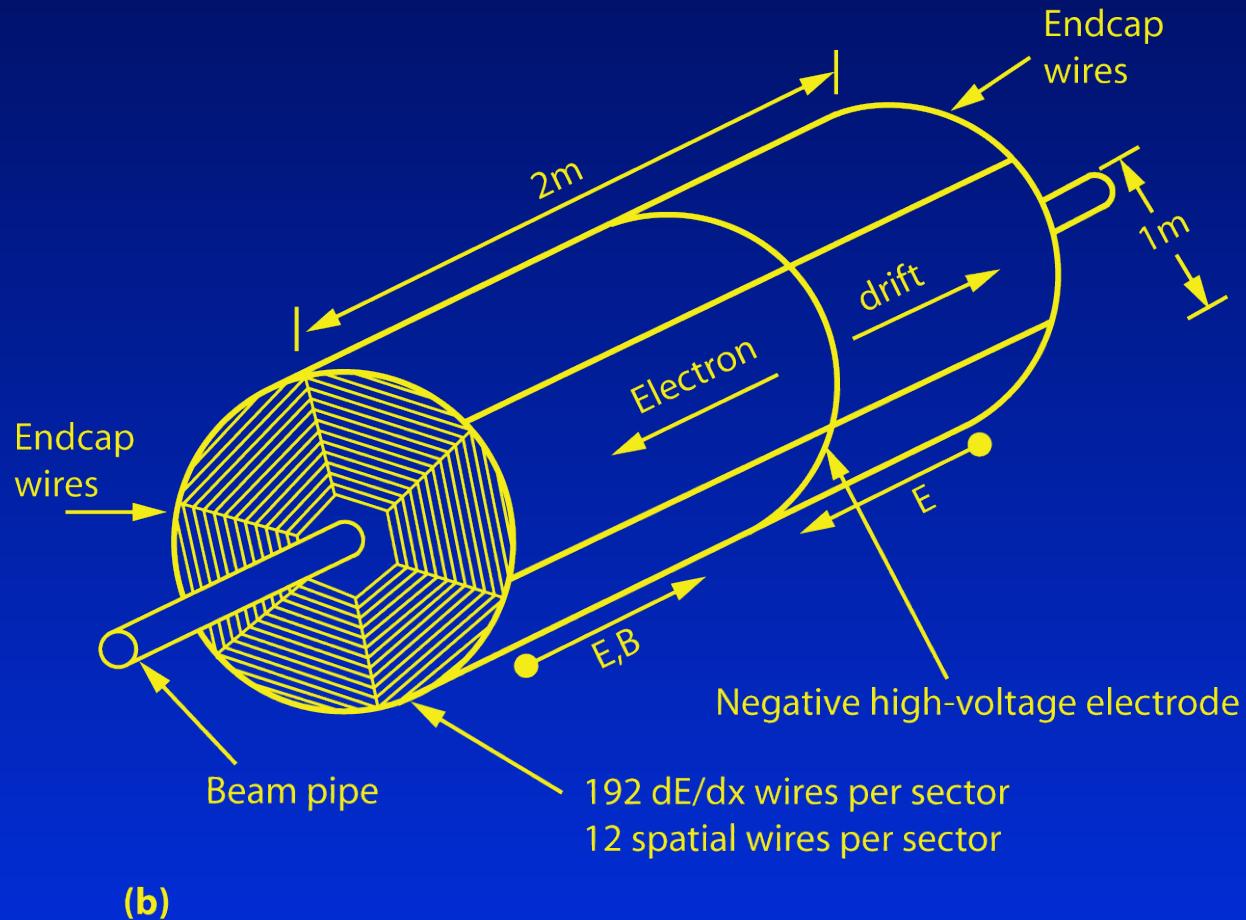
Beware: drift of electrons in the gas is influenced by magnetic field

$$v = \frac{e}{m_e} \frac{\tau}{1 + \omega^2 \tau^2} \left( \vec{E} + \frac{\omega \tau}{B} (\vec{E} \times \vec{B}) + \frac{\omega^2 \tau^2}{B^2} (\vec{E} \cdot \vec{B}) \vec{B} \right)$$

$$\omega = eB / m_e$$

If  $E//B$ , no influence on drift direction

# Time projection chamber



In modern version GEM or MICROMEGAS

## Wire chambers in brief

Simple and inexpensive devices > for large area

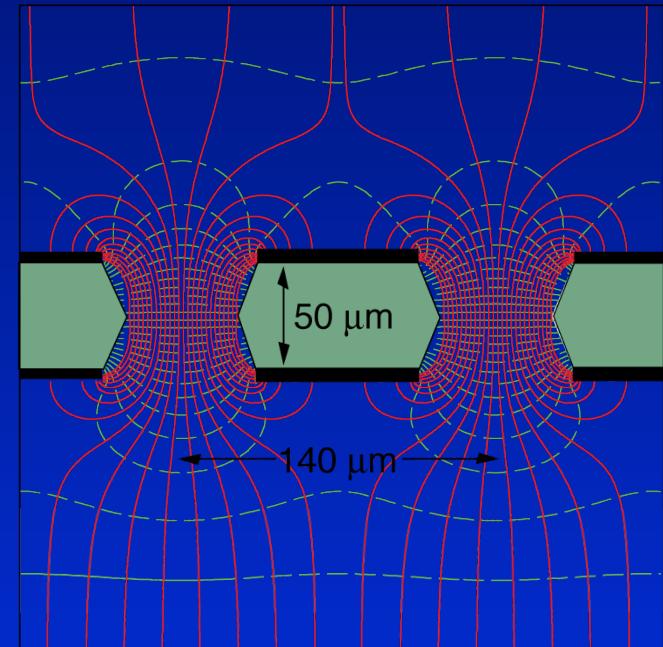
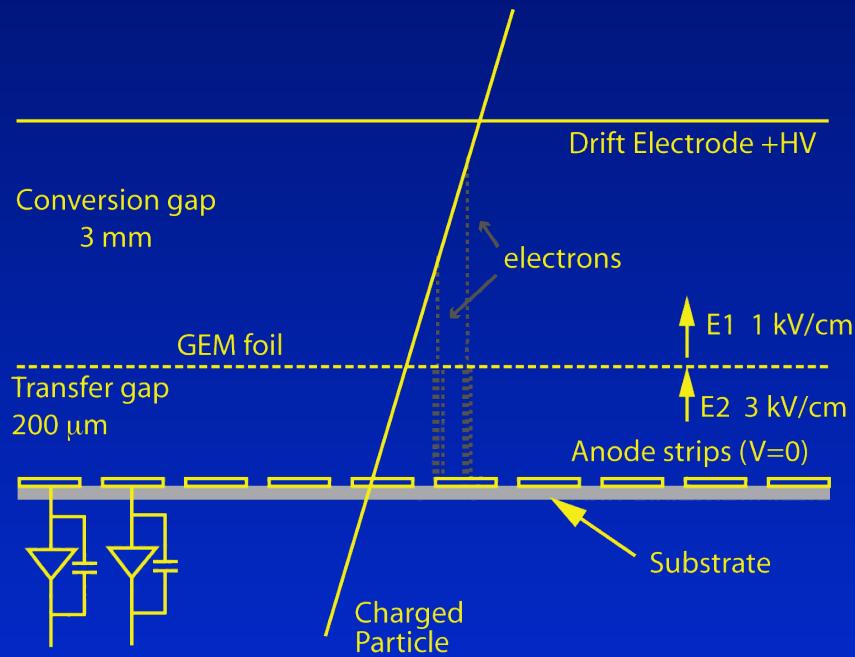
- need for constant gas flow
- aging: with proper care > few coulomb/cm wire
- the pulse shape is governed by the slow motion of the ions
- build-up of space charge from positive ions

Maximum Rate : (charge/track)  $\times$  count-rate  $\approx 10^{10}$  /mm/s

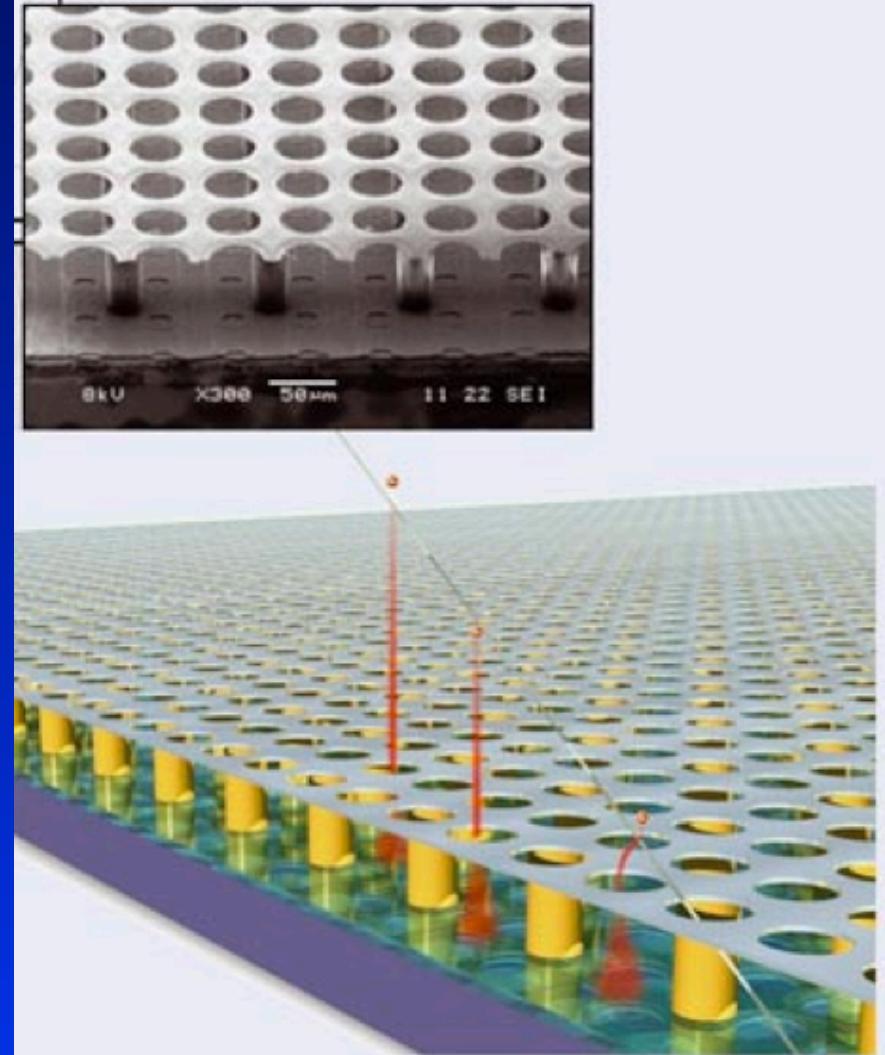
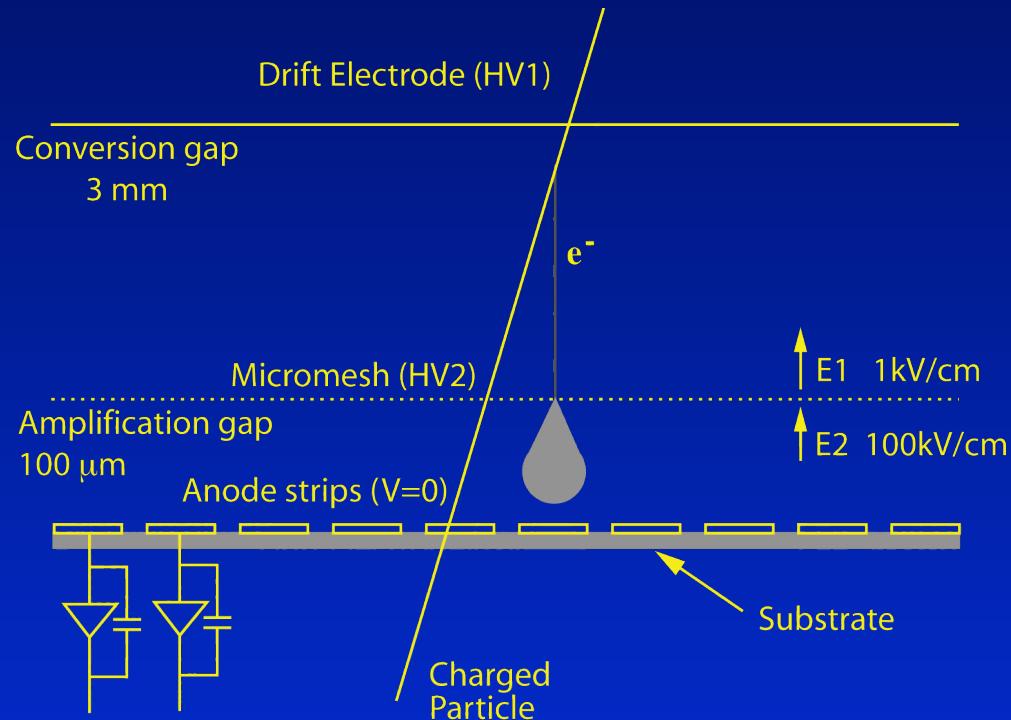
To overcome limitations related to slow ion drift:

Micropattern gas detectors

# Micropattern gas detectors: GEM counter

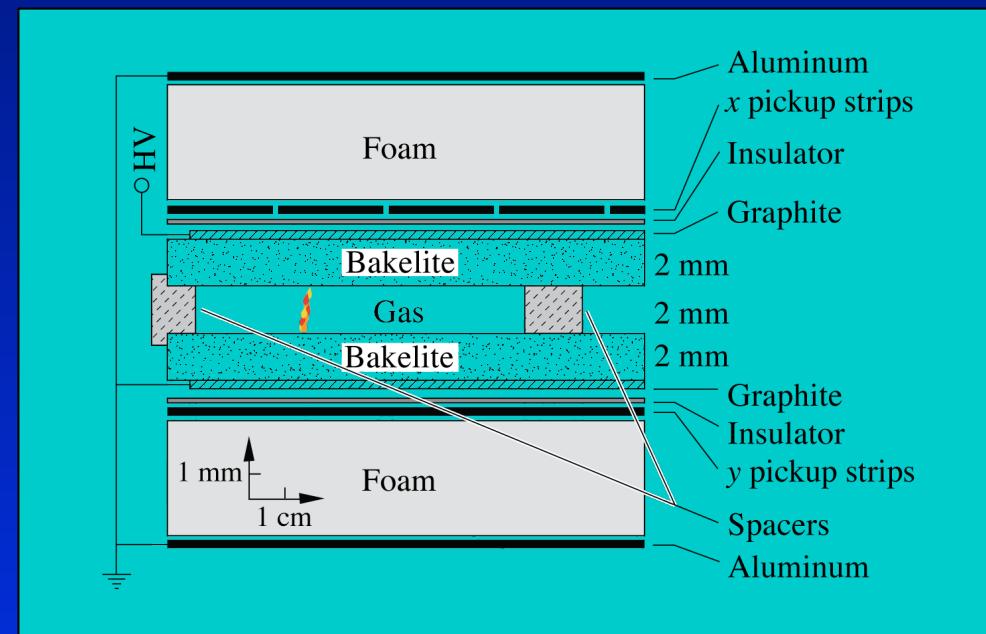
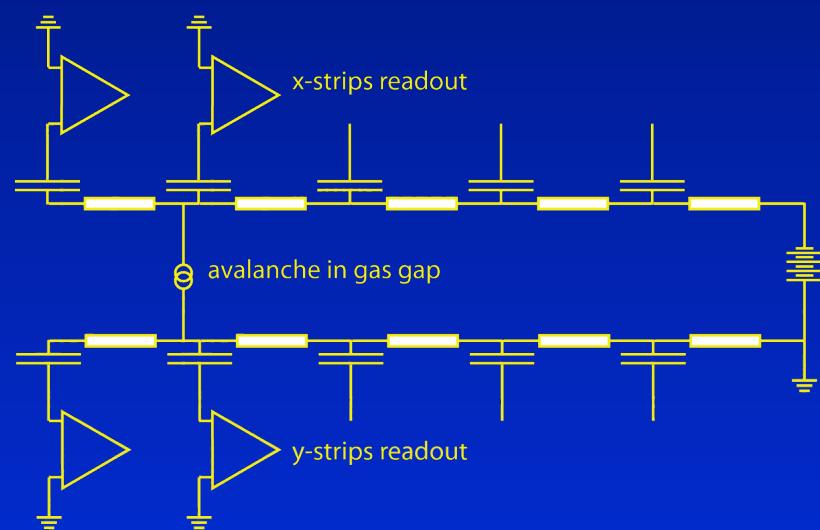


# Micropatter gas detectors: MICROMEGAS



GEM detectors and MICROMEGAS, have achieve good timing  $\approx 10$  ns, large rates  $\approx 25$  kHz/mm $^2$ , spatial resolution  $\approx 50$   $\mu$ m, and resistance to radiation comparable to silicon detectors.

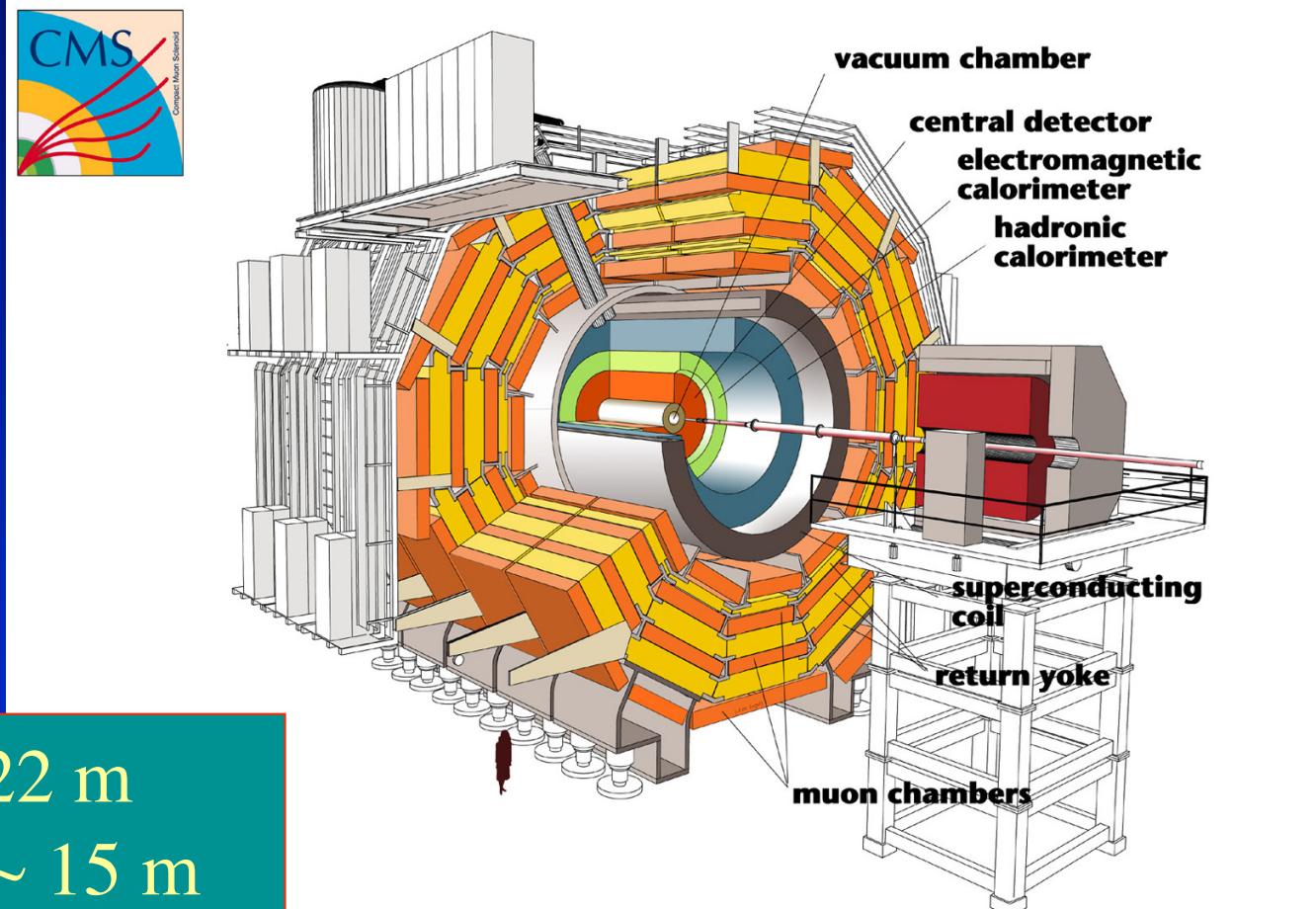
# Resistive plate chambers



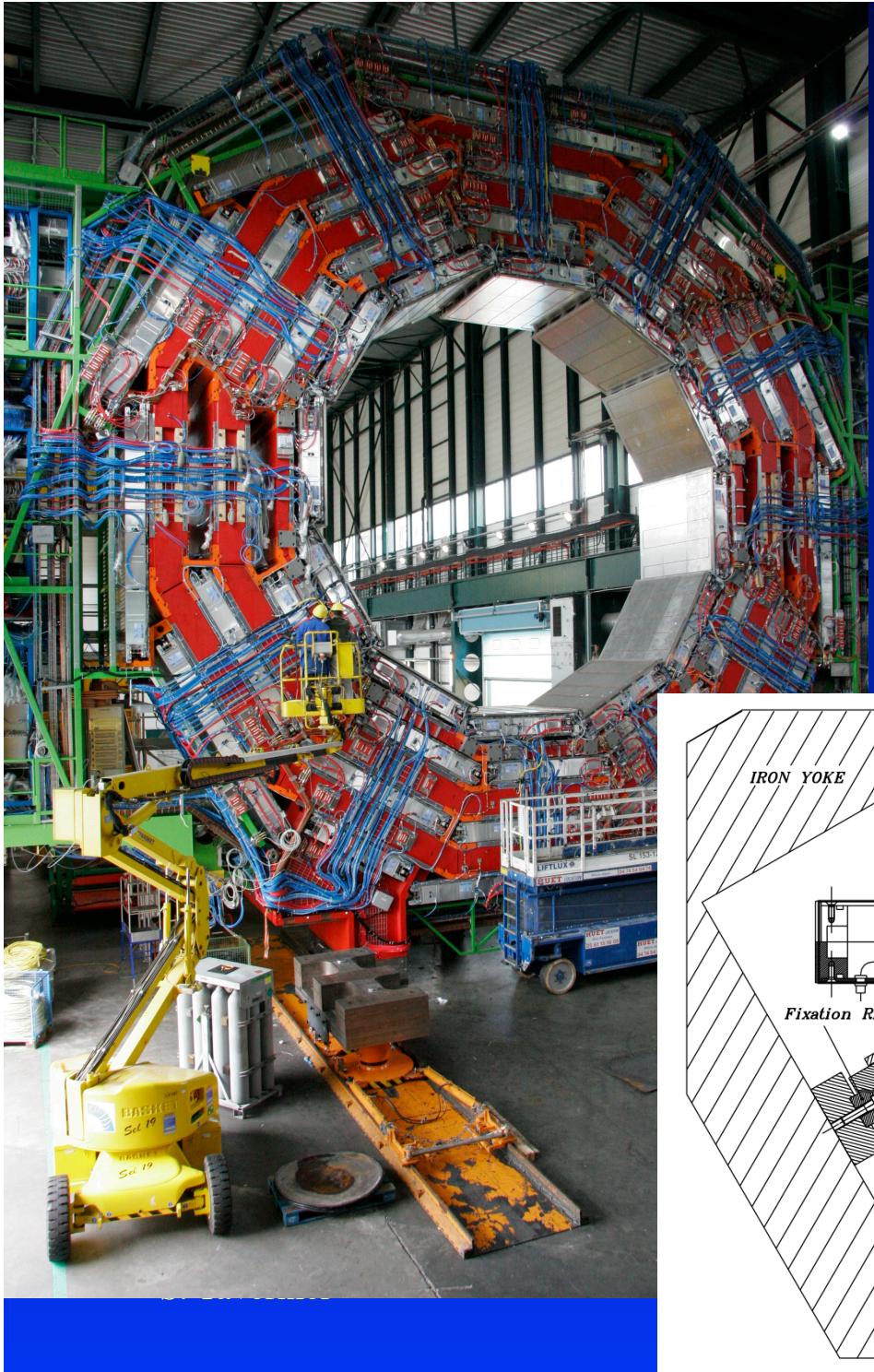
## Applications of detectors based on ionisation in gases

Mainly used for tracking of charged particles and when one does not need, or cannot afford, silicon tracking detectors, e.g. muon detection systems at high energy colliders.

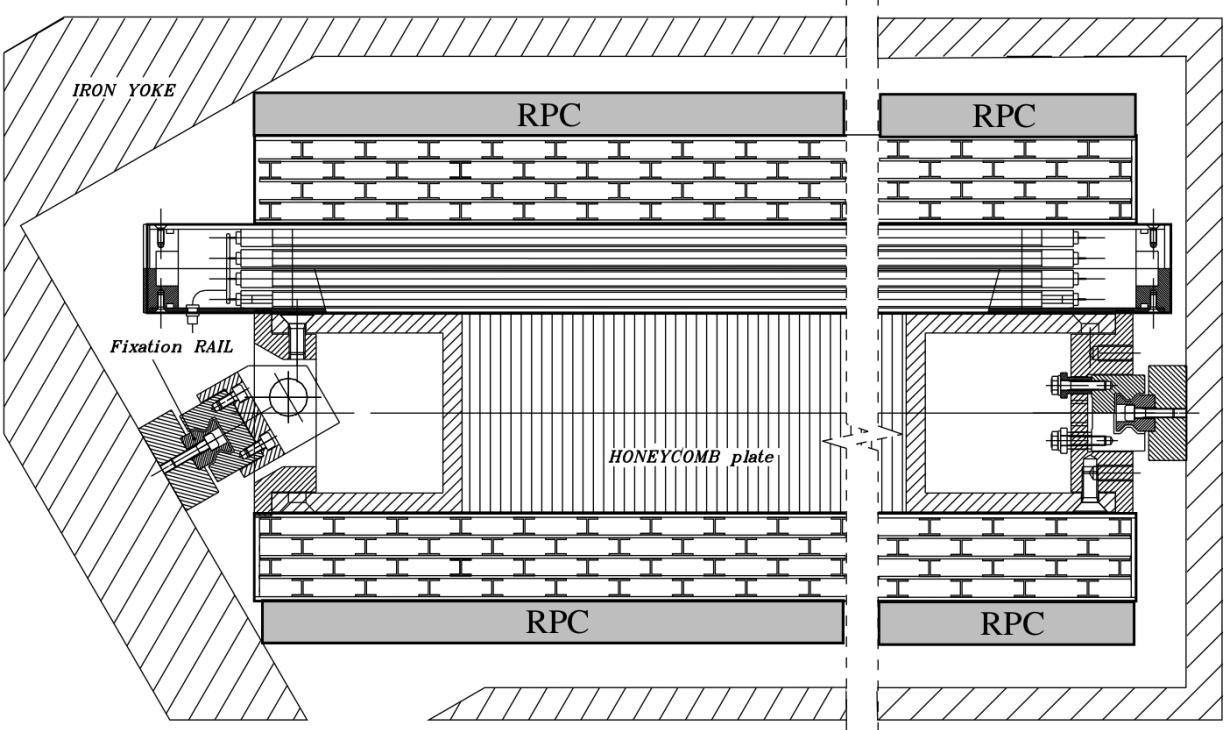
# Compact Muon Solenoid



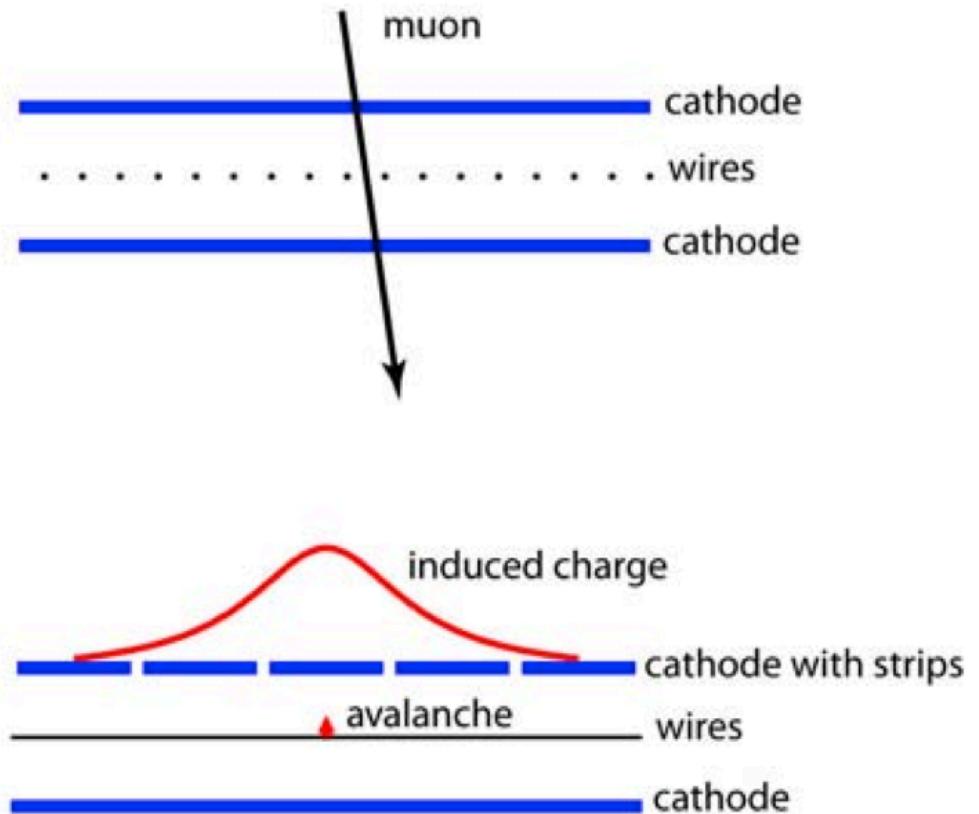
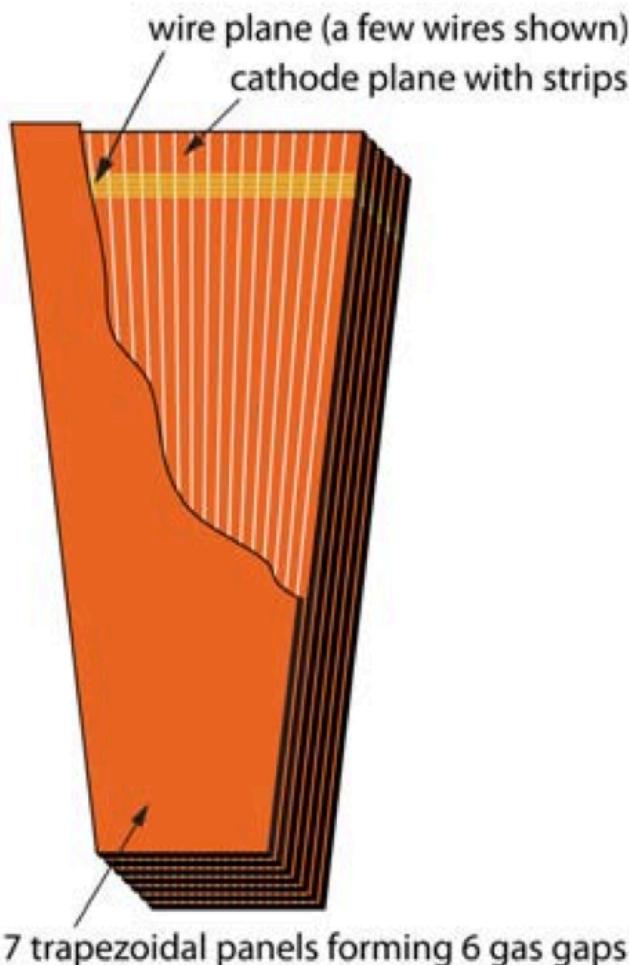
- ~ Length ~ 22 m
- ~ Diameter ~ 15 m
- ~ Weight ~ 14500 t



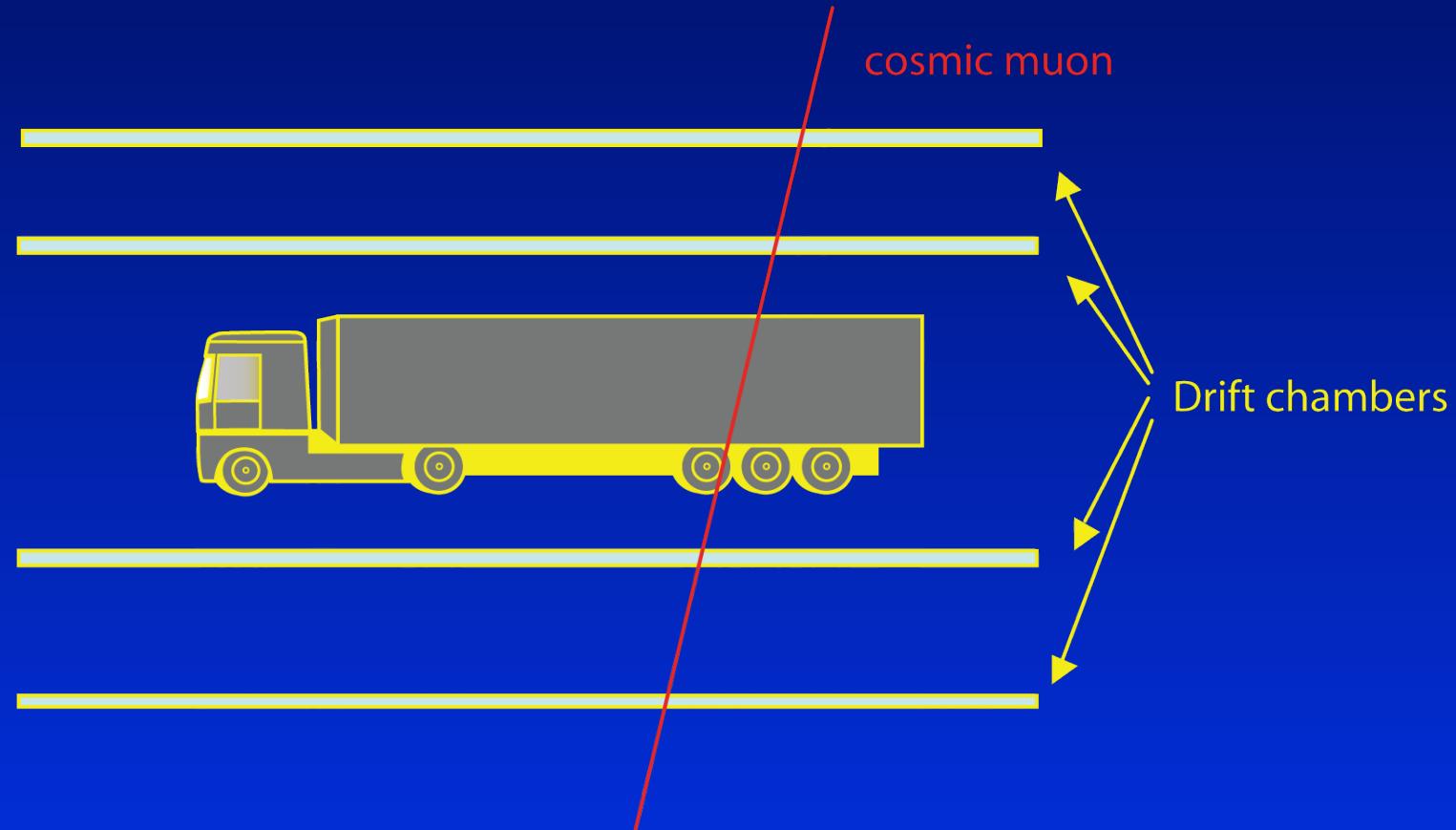
CMS: Muon system. Drift chambers in barrel.



# CMS: Muon system. Cathode strip chambers in barrel.



# Wire chambers in homeland security



*End of section on detectors based on ionisation in gases.*